

KNOWLEDGE ORGANISER

YEAR 8 – TERM 5



Think Like An
Environmentalist

Community, Collaboration and Challenge

ATTENDANCE MATTERS



EVERY DAY COUNTS

Missing just 1 day every 2 weeks is the same as missing 10% of the school year.

LEARNING

Being in school allows you the best opportunity to learn.



WELLBEING

Attending school supports your mental and emotional health.

FUTURE SUCCESS

Regular attendance at school is vital for building the key skills needed for future employment



EQUIPMENT



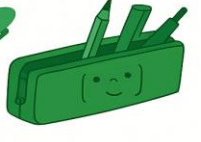
School Bag



Knowledge Organiser



Black and Green Pens



Pencil case



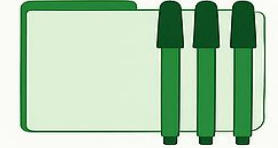
Calculator



Pencil



Rubber



Whiteboard and whiteboard pen



Highlighters



Ruler

SCHOOL DAY

9:00–9:05

AM Reg

9:05–10:20

Lesson 1

10:20–11:35

Lesson 2

11:35–12:05

Break 1

12:05–13:20

Lesson 3

13:20–13:50

Break 2

13:50–15:05

Lesson 4

15:05–15:30

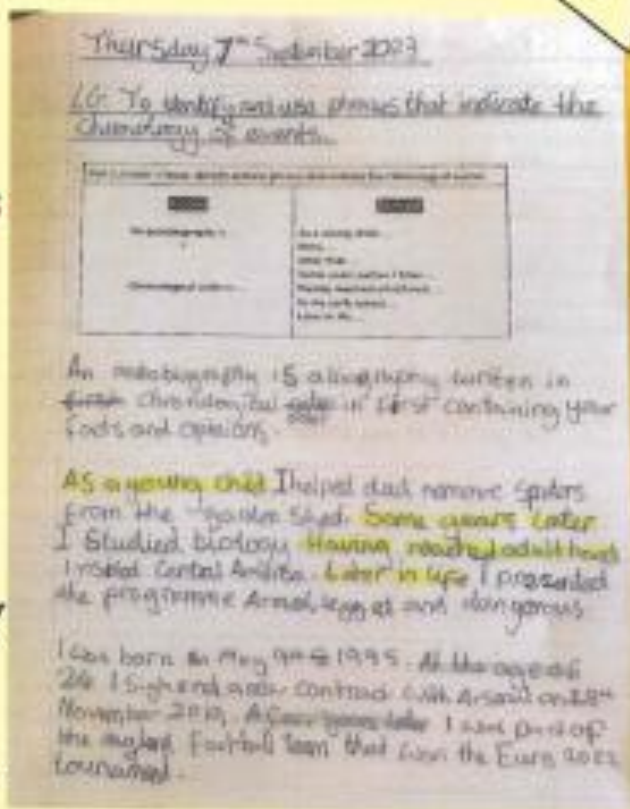
PM Reg – assembly or guided reading

Long date and learning goal (LG) on left and underlined.

Minimise errors

Use pencil or blue or black pen for all written work, ruler for straight lines.

Worksheets glued in straight, next to margin.



Numbered questions written in margin.

Always try your best to be neat (even when note taking).

Feel proud

Single line through mistake. No scribbles.

Prepare for the future

Respond to feedback in green pen.

Easier to read and give better feedback

Multiplication Grid

x	1	2	3	4	5	6	7	8	9	10	11	12
1	1	2	3	4	5	6	7	8	9	10	11	12
2	2	4	6	8	10	12	14	16	18	20	22	24
3	3	6	9	12	15	18	21	24	27	30	33	36
4	4	8	12	16	20	24	28	32	36	40	44	48
5	5	10	15	20	25	30	35	40	45	50	55	60
6	6	12	18	24	30	36	42	48	54	60	66	72
7	7	14	21	28	35	42	49	56	63	70	77	84
8	8	16	24	32	40	48	56	64	72	80	88	96
9	9	18	27	36	45	54	63	72	81	90	99	108
10	10	20	30	40	50	60	70	80	90	100	110	120
11	11	22	33	44	55	66	77	88	99	110	121	132
12	12	24	36	48	60	72	84	96	108	120	132	144

PERIODIC TABLE OF ELEMENTS

Chemical Group Block



1																	18							
1	1.0080															2	4.00260							
1	H Hydrogen Nonmetal															2	He Helium Noble Gas							
2	3	4	Atomic Number										13	14	15	16	17	18						
2	7.0	9.012183											5	10.81	6	12.011	7	14.007	8	15.999	9	18.9984...	10	20.180
2	Li Lithium Alkali Metal	Be Beryllium Alkaline Earth Me...											B Boron Metalloid	C Carbon Nonmetal	N Nitrogen Nonmetal	O Oxygen Nonmetal	F Fluorine Halogen	Ne Neon Noble Gas						
3	11	12	Name										13	14	15	16	17	18						
3	22.989...	24.305											13	26.981...	14	28.085	15	30.973...	16	32.07	17	35.45	18	39.9
3	Na Sodium Alkali Metal	Mg Magnesium Alkaline Earth Me...	Chemical Group Block										Al Aluminum Post-Transition M...	Si Silicon Metalloid	P Phosphorus Nonmetal	S Sulfur Nonmetal	Cl Chlorine Halogen	Ar Argon Noble Gas						
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36						
4	39.0983	40.08	44.95591	47.867	50.9415	51.996	54.93804	55.84	58.93319	58.693	63.55	65.4	69.723	72.63	74.92159	78.97	79.90	83.80						
4	K Potassium Alkali Metal	Ca Calcium Alkaline Earth Me...	Sc Scandium Transition Metal	Ti Titanium Transition Metal	V Vanadium Transition Metal	Cr Chromium Transition Metal	Mn Manganese Transition Metal	Fe Iron Transition Metal	Co Cobalt Transition Metal	Ni Nickel Transition Metal	Cu Copper Transition Metal	Zn Zinc Transition Metal	Ga Gallium Post-Transition M...	Ge Germanium Metalloid	As Arsenic Metalloid	Se Selenium Nonmetal	Br Bromine Halogen	Kr Krypton Noble Gas						
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54						
5	85.468	87.62	88.90584	91.22	92.90637	95.95	96.90636	101.1	102.9055	106.42	107.868	112.41	114.818	118.71	121.760	127.6	126.9045	131.29						
5	Rb Rubidium Alkali Metal	Sr Strontium Alkaline Earth Me...	Y Yttrium Transition Metal	Zr Zirconium Transition Metal	Nb Niobium Transition Metal	Mo Molybdenum Transition Metal	Tc Technetium Transition Metal	Ru Ruthenium Transition Metal	Rh Rhodium Transition Metal	Pd Palladium Transition Metal	Ag Silver Transition Metal	Cd Cadmium Transition Metal	In Indium Post-Transition M...	Sn Tin Post-Transition M...	Sb Antimony Metalloid	Te Tellurium Metalloid	I Iodine Halogen	Xe Xenon Noble Gas						
6	55	56	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86							
6	132.90...	137.33	178.49	180.9479	183.84	186.207	190.2	192.22	195.08	196.96...	200.59	204.383	207	208.98...	208.98...	209.98...	222.01...							
6	Cs Cesium Alkali Metal	Ba Barium Alkaline Earth Me...	Hf Hafnium Transition Metal	Ta Tantalum Transition Metal	W Tungsten Transition Metal	Re Rhenium Transition Metal	Os Osmium Transition Metal	Ir Iridium Transition Metal	Pt Platinum Transition Metal	Au Gold Transition Metal	Hg Mercury Transition Metal	Tl Thallium Post-Transition M...	Pb Lead Post-Transition M...	Bi Bismuth Post-Transition M...	Po Polonium Metalloid	At Astatine Halogen	Rn Radon Noble Gas							
7	87	88	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118							
7	223.01...	226.02...	267.1...	268.1...	269.1...	270.1...	269.1...	277.1...	282.1...	282.1...	286.1...	286.1...	290.1...	290.1...	293.2...	294.2...	295.2...							
7	Fr Francium Alkali Metal	Ra Radium Alkaline Earth Me...	Rf Rutherfordium Transition Metal	Db Dubnium Transition Metal	Sg Seaborgium Transition Metal	Bh Bohrium Transition Metal	Hs Hassium Transition Metal	Mt Meitnerium Transition Metal	Ds Darmstadtium Transition Metal	Rg Roentgenium Transition Metal	Cn Copernicium Transition Metal	Nh Nihonium Post-Transition M...	Fl Flerovium Post-Transition M...	Mc Moscovium Post-Transition M...	Lv Livermorium Post-Transition M...	Ts Tennessine Halogen	Og Oganesson Noble Gas							
	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71									
	138.9055	140.116	140.90...	144.24	144.91...	150.4	151.964	157.2	158.92...	162.500	164.93...	167.26	168.93...	173.05	174.9668									
	La Lanthanum Lanthanide	Ce Cerium Lanthanide	Pr Praseodymium Lanthanide	Nd Neodymium Lanthanide	Pm Promethium Lanthanide	Sm Samarium Lanthanide	Eu Europium Lanthanide	Gd Gadolinium Lanthanide	Tb Terbium Lanthanide	Dy Dysprosium Lanthanide	Ho Holmium Lanthanide	Er Erbium Lanthanide	Tm Thulium Lanthanide	Yb Ytterbium Lanthanide	Lu Lutetium Lanthanide									
	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103									
	227.02...	232.038	231.03...	238.0289	237.04...	244.06...	243.06...	247.07...	247.07...	251.07...	252.0830	257.0...	258.0...	259.1...	266.1...									
	Ac Actinium Actinide	Th Thorium Actinide	Pa Protactinium Actinide	U Uranium Actinide	Np Neptunium Actinide	Pu Plutonium Actinide	Am Americium Actinide	Cm Curium Actinide	Bk Berkelium Actinide	Cf Californium Actinide	Es Einsteinium Actinide	Fm Fermium Actinide	Md Mendelevium Actinide	No Nobelium Actinide	Lr Lawrencium Actinide									

01 Adjectives

THAT DESCRIBE:
age: young, old
colour: red, blue
condition: new, used
size: large, medium
speed: fast, slow
etc.

COMPARATIVE:
 smaller, better...

SUPERLATIVE:
 the smallest,
 the worst,
 the best...

08 Verbs

ACTION:
 to run, to organise,
 to read, to think...
 > Transitive
 or
 > Intransitive

LINKING:
 to be,
 to look, to appear,
 to seem, to smell...

**HELPING
 (= AUXILIARY):**
 can, may,
 will, must,
 should, to be,
 to have...

07 Pronouns

PERSONAL (subject):
 I, you, he, she, it,
 we,
 you, they

PERSONAL (object):
 me, you, him, her,
 it, us, you, them

PERSONAL (reflexive):
 myself, yourself,
 himself, herself,
 itself, ourselves,
 yourselves,
 themselves

DEMONSTRATIVE:
 this, these,
 that, those

POSSESSIVE:
 mine, yours, his,
 hers, its, ours,
 yours, theirs

INTERROGATIVE:
 how, where,
 when, which...?

INDEFINITE:
 somebody,
 anyone...

RELATIVE:
 that, which,
 whose, whom...

06 Prepositions

PLACE / DIRECTION:
 in, at, on,
 under, above,
 across,
 among,
 between...

TIME:
 in, at, on,
 over, until, about,
 during, before,
 after, while,
 through...

**OTHER (agent,
 phrase...):**
 by, with, on, over,
 to, up, within,
 beyond, for...

05 Nouns

COMMON NOUNS: house, dog, laptop...

PROPER NOUNS:
 (Capitalised)
 London, Paris,
 James, William,
 Julia, Jennifer...

> **VERBAL:** swimming...

> **COLLECTIVE:** choir, jury...

> **COMPOUND:** mother-in-law...

> **COUNTABLE:** book, day...

> **UNCOUNTABLE:** traffic, calm...

> **ABSTRACT V. CONCRETE:** wit vs. road...

02 Adverbs

PLACE:
 here, there,
 outside, everywhere,
 upstairs, nowhere,
 somewhere....

TIME:
 ago, before, since,
 yet, for, still,
 afterwards...

FREQUENCY:
 often, never,
 sometimes, always

MANNER:
 just, quite,
 quickly, hardly,
 well, carefully,
 barely, almost,
 scarcely,
 beautifully...

03 Conjunctions

COORDINATING:
 and, or, but,
 yet, nor, for, so

CORRELATIVE:
 both... and...,
 either... or...,
 just as... so...,
 whether... or...,
 neither... nor...,
 not only... but also...

SUBORDINATING:
 after, since, if,
 while, although,
 before, because,
 unless

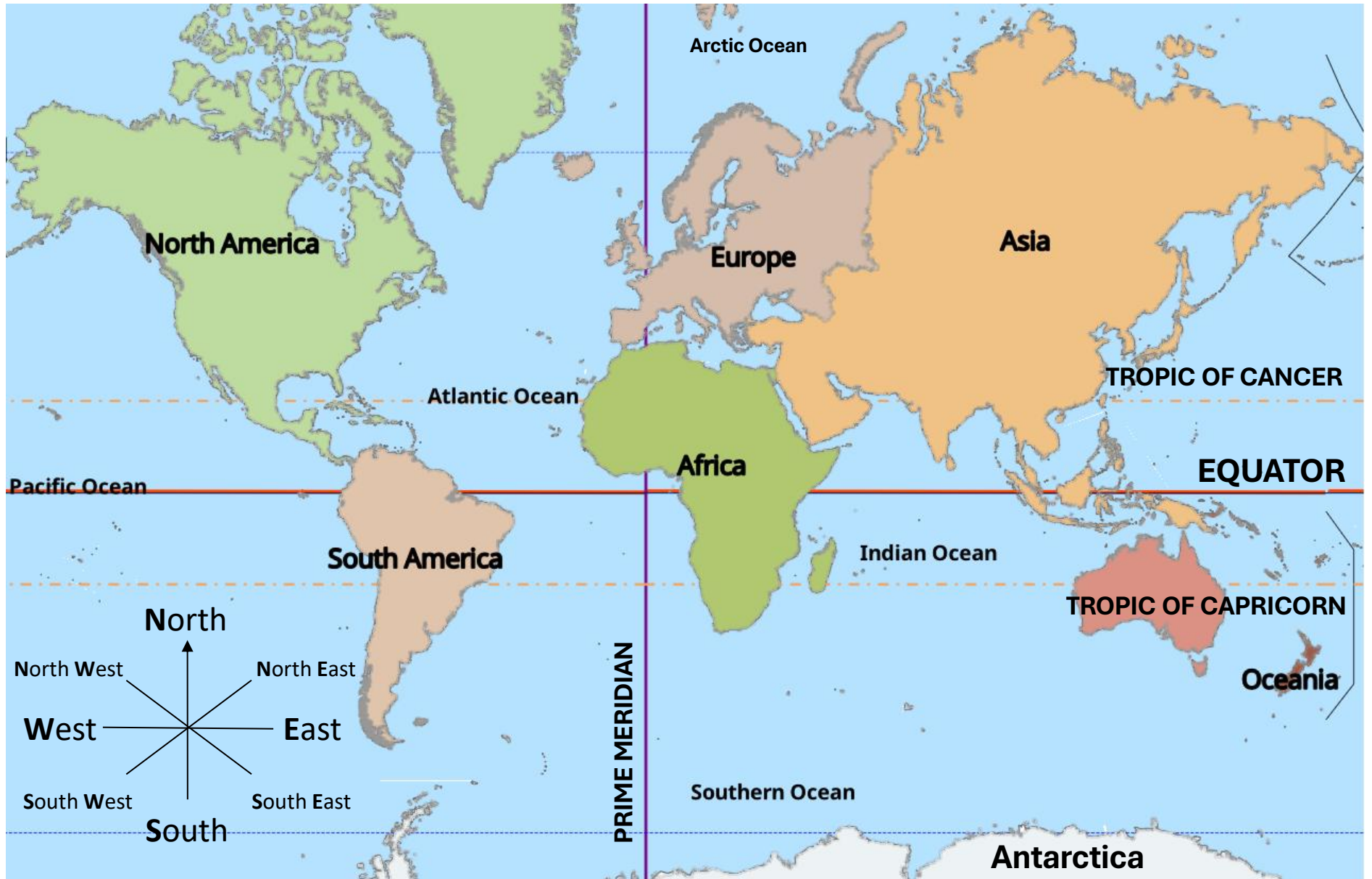
04 Determiners

TELLS US WHICH:
 each, every,
 some, none,
 all...

TELLS US WHOSE:
 my, your, her, his, its,
 our, your, their (= possessive
 adjectives or determiners)



World Map



Year 8 Ceramics

Art

Term 5

Clay Coil Pot Lesson Plan

Theme: Exploring Form and Function through Clay Coil Pots

Focus: Building practical skills in ceramics, understanding historical and cultural contexts, and developing creative design.



Building Coils

- **Objective:** Practice coil rolling and beginning to build the pot.
- **Activities:**
- Clay safety and etiquette briefing.
- Demonstration: Rolling and attaching a base slab.
- Students create the base for their pots.
- Demonstration: Even coil rolling and slip-and-score technique.
- Begin building: 3–4 coils attached to the base.
- Emphasis on neat joins and structural stability.

Building Coils

- **Objective:** Continue building, experimenting with shape and height.
- **Activities:**
 - Demonstration: Controlling form (inward/outward curves).
 - Continue building up walls; apply some decorative textures using found materials.
- Design 2 or more pattern ideas for surface decoration, creating tiny holes using straws and stamps.

Surface Texture & Decoration

- **Objective:** Add texture or relief designs to the pot.
- **Activities:**
- Demonstration: Texture tools, incising, impressing, relief.
- Add patterns, textures, and details.
- Finish building – refine edges and check for cracks.



Drying & Bisque Firing

- **Objective:** Understand the firing process.
- **Activities:**
 - Prepare pieces for drying and firing.
 - Presentation: Kilns, firing stages, glaze vs. paint.
 - Begin colour planning in sketchbook – sample colour swatches and test schemes.

Decorating – Underglaze or Paint

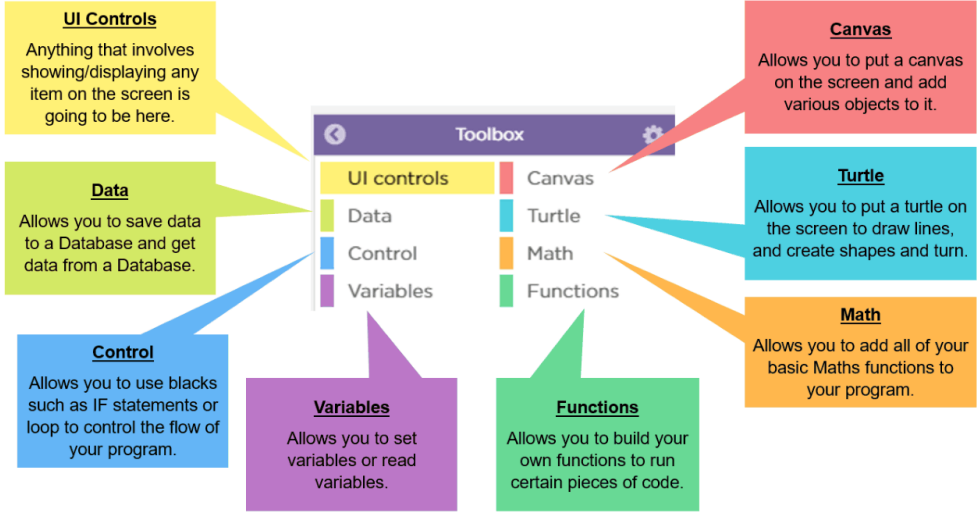
- **Objective:** Apply colour using appropriate ceramic materials.
- **Activities:**
- Demonstration: Glazing or painting technique.
- Apply finish to bisque-fired pots (students can choose glaze, acrylic, or both).



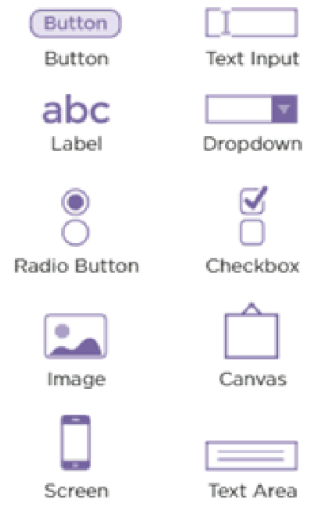
Computing

App Lab Knowledge Organiser

App Lab Toolbox Guide



Design Elements



What is event-driven programming?

Event-driven programming is an approach to programming in which code is written to respond to events. Events can be triggered by users, such as clicking on a button or entering some text. In automated systems, sensors can be used to detect events such as when a particular temperature is reached in a glasshouse or a specific water level is detected on a flood defence system.

Drop down boxes are usually followed by the 'change' event.

Buttons are usually followed by the 'click' event.

IF statements can help set conditions to tell the program what to do.

```
onEvent (▼ "checkAnswer", ▼ "click", function () {
  if (getChecked("radio_button4")) {
    setScreen (▼ "correctScreen");
  } else {
    setScreen (▼ "wrongScreen");
  }
});
```

```
onEvent (▼ "DropdownDoctor", ▼ "change", function () {
  2 if (getText (▼ "DropdownDoctor") == "David Tennant") {
    setImageURL (▼ "imageDoctor", ▼ "david-tennant2.png");
  } else {
    4 setImageURL (▼ "imageDoctor", ▼ "matt-smith2.png");
    5
    6
    7
  }
});
```

Year 8

Drama

Term 5

Script Writing Techniques

Act One Scene 1 - This is the start of the play. An act is a way of dividing up a play (a bit like a chapter in a story book). An act can be made up of different scenes.

Scene - When you are writing a script, you start a new scene when the location, scenery or time changes.

Script - The script has information that tells everyone where the action takes place. It gives more detail about the scene and which characters are going to be in it:

Scene 3 - A cave in the middle of the desert.

A young boy called Aladdin is in a cave, surrounded by piles of treasure. Nearby there is a lamp.

Dialogue - When a character speaks in a play, we usually see their name in capitals. This is followed by the dialogue that the character is to say. Script writers can also get actors to perform the dialogue in a particular way. They do this by adding the direction in brackets:

ALADDIN: *(whispered)* I've found the magic lamp!

ALADDIN walks over to the lamp. He picks it up and rubs it. In a puff of smoke, a magical genie appears in front of him.

The script gives the director and actors all the words and instructions to tell a story in just they way the writer wishes.

Terminology

Genre - a style or category of art, music or literature

Dialogue - a conversation between two or more people as a feature of a book, play, or film.

Adaptation - changing from one genre to another

Silent Film - action without any words

Proxemics - the use of space between characters on a stage to communicate their relationships, status, and situation

Stimulus - a starting point or a source of inspiration.

Children's Theatre Techniques

Mime - the art of expressing ideas, emotions, and stories without using words.

Narration - the story told directly to the audience.

Sound effects - noises made to enhance the action.

Characterisation - the process an actor uses to develop a complete and believable character

Exaggeration - overstating or overemphasising a situation, emotion, or character for effect.

Adaptations to enjoy

Matilda (Book, play and film)

The Hunger Games (Book, play and film)

Under Milk Wood (radio, play and film)

My neighbour Totoro (film and play)

Adapting for Stage

English

CONTEXT (Elizabethan England & Verona)

Patriarchal Society: Fathers controlled daughters' marriages. Women were expected to obey and remain chaste. Juliet's defiance is shocking for the time.

Honour & Reputation: Family honour was vital. Insults led to duels and revenge. The feud reflects this obsession.
Religion: Marriage was sacred. Friar Lawrence's secret ceremony shows conflict between faith and love.

Marriage & Social Status: Wealthy families married for money, power, and alliances. Juliet is treated as a bargaining tool by her father.

Fate & The Plague: Fate is a driving force ('star-crossed lovers'). The plague quarantine prevents the vital message reaching Romeo.

Tragic Hero : Romeo's impulsiveness and emotional intensity lead to catastrophe.

Setting – Verona: A city of wealth and rivalry, mirroring Elizabethan fears of disorder and social conflict.

THEMES (Key Ideas & Motifs)

- **Love & Impulsiveness:** Romeo & Juliet fall in love at first sight. Their love is genuine but reckless and leads to destruction.
- **Family & Honour:** Honour shapes every decision. Characters cannot ignore insults. Violence escalates until the lovers' deaths force peace.
- **Women & Power:** Women have limited influence. Juliet is seen as her father's property. She values virtue and chastity but also asserts independence.
- **Marriage & Duty:** Arranged marriages were common. Juliet is torn between duty and personal happiness. Romeo & Juliet reject family expectations.
- **Fate & Luck:** Fate is introduced in the prologue. Chance events (the servant inviting Romeo, delayed message, Paris at the tomb) drive the tragedy.
- **Religion & Morality:** Religion shapes behaviour and expectations. Friar Lawrence tries to help but contributes to the tragedy.
- **Youth & Conflict:** Rash decisions and emotional extremes show the destructive power of youthful passion.
- **Death & Reconciliation:** Death becomes the only way to unite love and end hatred.

PLOT (Act-by-Act Summary)

Prologue: Introduces the Montague-Capulet feud. Warns the audience that the lovers will die because of their families' hatred.

Act 1: Romeo crashes the Capulet ball and meets Juliet. They fall in love instantly despite their families' hatred.

Act 2: Romeo and Juliet secretly marry with Friar Lawrence's help, hoping their union will end the feud.

Act 3: Tybalt kills Mercutio. Romeo kills Tybalt in revenge and is banished. Juliet is devastated but remains loyal to Romeo.

Act 4: Juliet's parents arrange her marriage to Paris. She takes a sleeping potion to avoid the marriage.

Act 5: Romeo, misinformed, believes Juliet dead and drinks poison. Juliet wakes, sees Romeo dead, and kills herself.

Resolution: The families reconcile, realising their hatred caused the tragedy.

1. Russia's Location



3. Russia's Physical Characteristics



Key

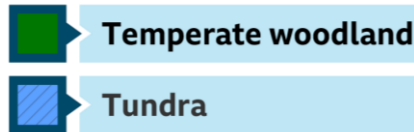


Russia is the largest country in the world. It covers an area of 17,098,242 km², which is around 70 times greater the size of the UK. The country spans 11 time zones.

Russia contains several biomes examples of which include:

Taiga: vegetation covers most of the country. This is Evergreen forest, with trees such as spruce, fir and pine.

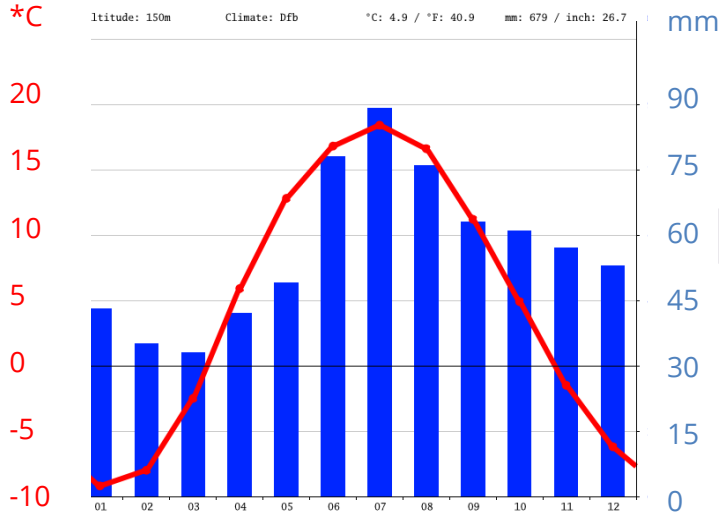
Tundra: biomes are found in the far north of Russia. Small shrubs, mosses and lichens grow, and these provide food for animals such as reindeer. Larger species of vegetation struggle to exist in the cold, dry conditions found here.



Temperate woodland and steppe: are found in the south of Russia. These are warmer landscapes consisting of deciduous trees and grasses.

Deserts: located in the far south of Russia. These areas have very little precipitation and so few plants can grow here.

2. Russia's Climate - Moscow



Population distribution:

Most people live in the west of the country. This is where the capital city of Moscow and is located, as well as many other larger cities, such as St Petersburg and Kazan. Around 75 per cent of Russia's population live in cities, where there are jobs and opportunities. Few people live in the far north, where temperatures are very low.

4. Russia's Human Characteristics

Life expectancy and literacy rates:

With approximately 144 million people, Russia is the ninth most populous country in the world. Life expectancy is almost 73 years, so the average person in Russia can expect to live beyond retirement age. Literacy rates are 100 per cent. This means that all Russian people can read and write.

St Petersburg



History

Knowledge Organiser – HISTORY

Topic: WW2



sKey Events in World War 2	
September 1 st 1939	Germany invades Poland.
September 3 rd 1939	Britain and France declare war on Germany (Start of WW2).
January, 1940	Rationing introduced across the UK.
May – June, 1940	Dunkirk evacuated and France surrenders to Germany. Germany uses blitzkrieg to take over much of Western Europe.
July, 1940	Germany launches air attacks on Great Britain (The Battle of Britain and the Blitz begin) Germany, Italy and Japan sign the Tripartite Pact creating the axis alliance.
December 7 th 1941	The Japanese attack the US navy in Pearl Harbour. The next day, the USA enters the war fighting with the allies
June 6 th 1944	D-day and the Normandy invasion. Allied forces invade France and push back the Germans.
April 30 th 1945	Adolf Hitler commits suicide
May 7 th 1945	Germany surrenders and Victory in Europe is declared the next day.
August 1945	Atomic bombs dropped on Hiroshima and Nagasaki, Japan by the USA killing approximately 226,000 people.
September 2 nd 1945	Japan surrenders signalling the end of WW2
July 1954	Rationing ends in the UK

Leaders

United Kingdom



Born: 30th Nov 1874
Died: 24th Jan 1965

"I am easily satisfied with the very best."
Winston Churchill

Germany



Born: 20th April 1889
Died: 30th April 1945

"Strength lies not in defence but in attack."
Adolf Hitler

Countries involved in WW2

Allies

USA, Britain, France, USSR, Australia, Belgium, Brazil, Canada, China, Denmark, Greece, Netherlands, New Zealand, Norway, Poland, Czechoslovakia, Yugoslavia, India, South Africa

Axis

Germany, Italy, Japan, Hungary, Romania, Bulgaria, Finland

Key Vocabulary

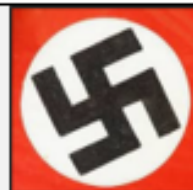
Allies	Countries which fought on Britain's side.	Axis	Countries which fought on Germany's side.
Evacuee	Someone who was evacuated. Moved from a dangerous place to be safer.	Nazi	Member of a fascist German political party. They came to power in 1933.
Black out	Ensuring no lights were visible after dark so buildings can't be spotted.	Blitz	Aerial bombings on the UK, mainly cities including London, by Germany.
Rationing	The controlled distribution of limited resources such as food and clothing.	Holocaust	Mass murder of Jewish people and other groups by the Nazis.
Air raid shelter	A building to protect people from bombs. Made from corrugated iron usually found in people's gardens.	Fascism	Right wing political view associated with not allowing opposition and total control by a dictator.
Trenches	A long narrow ditch used for troops to shelter from enemy attack.	Blitzkrieg	Translates to 'lightning war'. Germany's quick strike invasion of Western Europe.



Enigma, machine used by Nazis to communicate



Children being evacuated out of London



Swastika, the Nazi party symbol



Supermarine Spitfire Mk 1, British fighter plane

Angles in parallel lines and polygons

@whisto_maths

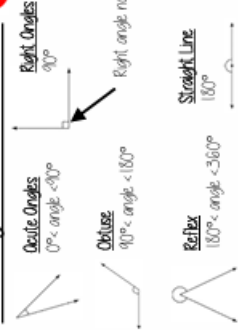
What do I need to be able to do?

- Identify alternate angles
- Identify corresponding angles
- Identify co-interior angles
- Find the sum of interior angles in polygons
- Find the sum of exterior angles in polygons
- Find interior angles in regular polygons

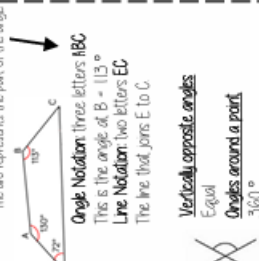
Keywords

- Parallel:** Straight lines that never meet
- Angle:** The figure formed by two straight lines meeting (measured in degrees)
- Transversal:** A line that cuts across two or more other (normally parallel) lines
- Isosceles:** Two equal size lines and equal size angles (in a triangle or trapezium)
- Polygon:** A 2D shape made with straight lines
- Sum:** Addition (total of all the interior angles added together)
- Regular polygon:** All the sides have equal length, all the interior angles have equal size

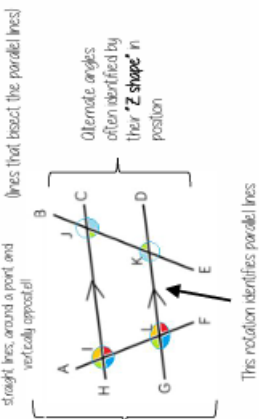
Basic angle rules and notation



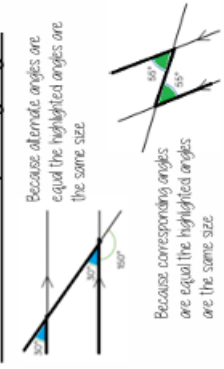
Angle Notation



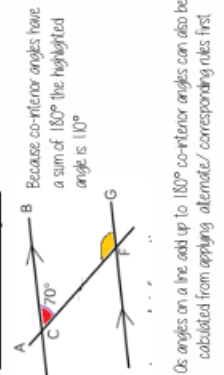
Parallel lines



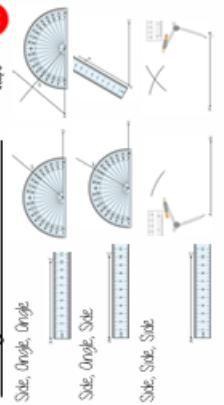
Alternate/ Corresponding angles



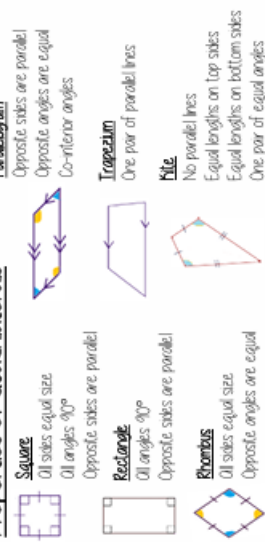
Co-interior angles



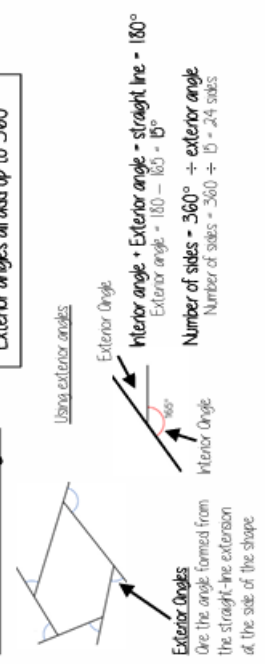
Triangles & Quadrilaterals



Properties of Quadrilaterals

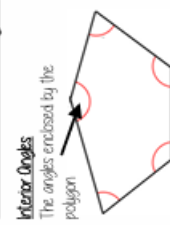


Sum of exterior angles

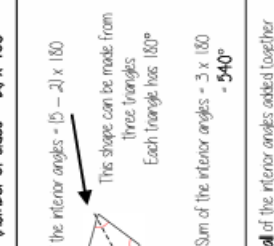


Exterior angles all add up to 360°

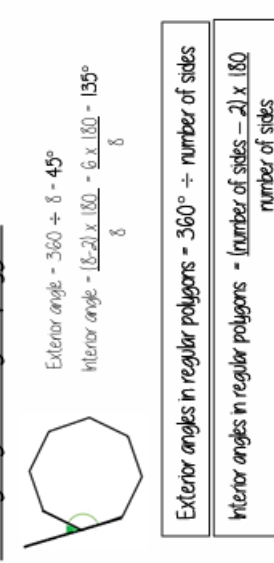
Sum of interior angles



number of sides - 2 x 180



Missing angles in regular polygons



Area of trapezia and Circles

@whisto_maths

What do I need to be able to do?

- By the end of this unit you should be able to:
 - Recall area of basic 2D shapes
 - Find the area of a trapezium
 - Find the area of a circle
 - Find the area of compound shapes
 - Find the perimeter of compound shapes

Keywords

- Congruent:** The same
- Area:** Space inside a 2D object
- Perimeter:** Length around the outside of a 2D object
- π (π):** The ratio of a circle's circumference to its diameter
- Perpendicular:** At an angle of 90° to a given surface
- Formula:** A mathematical relationship/ rule given in symbols. E.g. $b \times h = \text{area of rectangle/ square}$
- Unitarily (∞):** A number without a given ending (too great to count to the end of the number) — never ends
- Sector:** A part of the circle enclosed by two radii and an arc

Area — rectangles, triangles, parallelograms

Rectangle



Base x Height

Parallelogram/ Rhombus



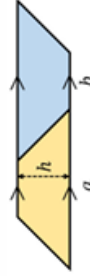
Base x Perpendicular height

Area of a trapezium

$$\frac{(a + b) \times h}{2}$$

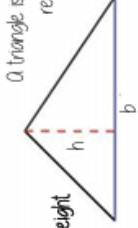


Why?



- Two congruent trapeziums make a parallelogram
- New length $(a + b)$ x height
- Divide by 2 to find area of one.

Triangle

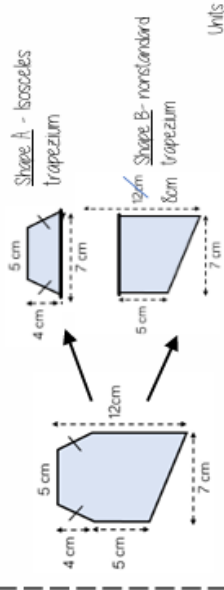


$\frac{1}{2} \times \text{Base} \times \text{Perpendicular height}$

A triangle is half the size of the rectangle it would fit in

Compound shapes

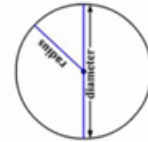
To find the area compound shapes often need splitting into more manageable shapes first. Identify the shapes and missing sides etc. first



Shape A + Shape B = total area

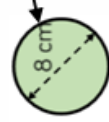
$$\frac{(5 + 7) \times 4}{2} + \frac{(5 + 7) \times 5}{2} = 24 + 45.5 = 69.5 \text{ cm}^2$$

Units



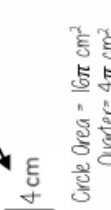
Area of a circle
 $\pi \times \text{radius}^2$

Read the question — leave in terms of π or if $\pi \approx 3$ (provides an estimate for answers)



Diameter = 8cm
 \therefore Radius = 4cm

$$\begin{aligned} \pi \times \text{radius}^2 &= \pi \times 4^2 \\ &= \pi \times 16 \\ &= 16\pi \text{ cm}^2 \end{aligned}$$



Circle Area = $16\pi \text{ cm}^2$
Quarter = $4\pi \text{ cm}^2$

Area of a circle (Calculator)



SHIFT $\times 10^{\square}$

How to get π symbol on the calculator

It is important to round your answer suitably — to significant figures or decimal places. This will give you a decimal solution that will go on forever!

Compound shapes including circles

Compound shapes are not always area questions. For Perimeter you will need to use the circumference

$$\text{Circumference} = \pi \times \text{diameter}$$

Spotting diameters and radii



This dimension is also the diameter of the semi circles.

$$\begin{aligned} \text{Arc lengths} &= \pi \times 64 \\ &= 64\pi \end{aligned}$$

Don't need to halve this because there are 2 ends which make the whole circle

Arc lengths + Straight lengths = total perimeter

$$\begin{aligned} &= 64\pi + 150 + 150 \\ &= \underline{300 + 64\pi} \text{ m} \\ \text{OR} &= \underline{501.1 \text{ m}} \end{aligned}$$

Still remember to split up the compound shape into smaller more manageable individual shapes first

Line symmetry and reflection

What do I need to be able to do?

By the end of this unit you should be able to:

- Recognise line symmetry
- Reflect in a horizontal line
- Reflect in a vertical line
- Reflect in a diagonal line

Keywords

Mirror line: a line that passes through the center of a shape with a mirror image on either side of the line.

Line of symmetry: same definition as the mirror line.

Reflect: mapping of one object from one position to another of equal distance from a given line.

Vertex: a point where two or more line segments meet.

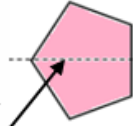
Perpendicular: lines that cross at 90° .

Horizontal: a straight line from left to right (parallel to the x axis)

Vertical: a straight line from top to bottom (parallel to the y axis)

Lines of symmetry

Mirror line (line of reflection)



Rhombus
Two lines of symmetry

Parallelogram
No lines of symmetry



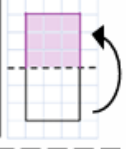
Shapes can have more than one line of symmetry...

This regular polygon (a regular pentagon) has 5 lines of symmetry

A circle has an infinite amount of lines of symmetry

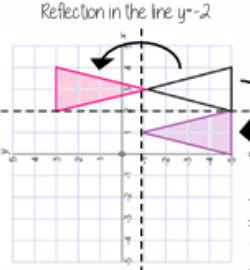


Reflect horizontally/ vertically (1)



Reflection in a vertical line

Reflection on an axis and

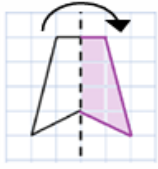


Reflection in the line $x=2$

Reflection in the line $y=-2$

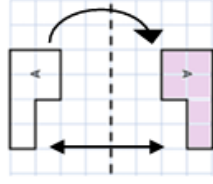
Note: a reflection doubles the area of the original shape

Reflection in a horizontal line

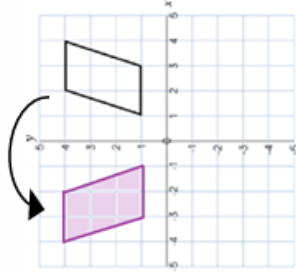


Reflect horizontally/ vertically (2)

All points need to be the same distance away from the line of reflection

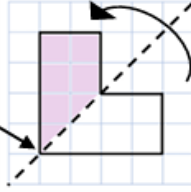


Reflection in the line y axis - this is also a reflection in the line $x=0$

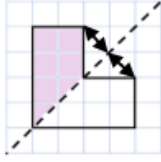


Reflect Diagonally (1)

Points on the mirror line don't change position

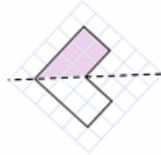


Fold along the line of symmetry to check the direction of the reflection



Turn your image

If you turn your image t , becomes a vertical/ horizontal reflection (also good to check your answer this way)

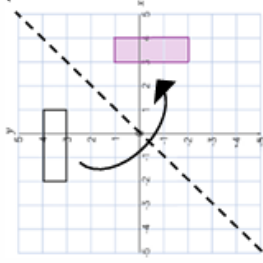


Drawing perpendicular lines

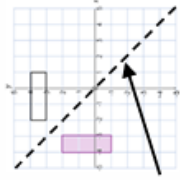
Perpendicular lines to and from the mirror line can help you to plot diagonal reflections

Reflect Diagonally (2)

This is the line $y = x$ (every y coordinate is the same as the x coordinate along this line)



This is the line $y = -x$
The x and y coordinate have the same value but opposite sign



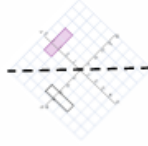
Lines parallel to the x and y axis

REMEMBER

Lines parallel to the x -axis are $y = \dots$
Lines parallel to the y -axis are $x = \dots$

Turn your image

If you turn your image t , becomes a vertical/ horizontal reflection (also good to check your answer this way)



Year 8


Music

Term 5

Year 8: Song writing – Playing / Analysing.

Rock – Chords and lyrics – Use Keyboard phone App


D
She's got a smile that it seems to me
C
Reminds me of childhood memories
G
Where everything
D
Was as fresh as a bright blue sky
D
Now and then when I see her face
C
She takes me away to that special place
G
And if I stared too long
D
I'd probably break down and cry
[Chorus]
A C D
Whoa oh, sweet child o' mine
A C D
Whoa, oh, oh, oh sweet love o' mine



Ballad – Chords and lyrics –

Use Keyboard phone App

G D
When the rain is blowing in your face
F C
And the whole world is on your case
Cm G
I could offer you a warm embrace
A7 D7 G
To make you feel my love
G D
When evening shadows and the stars appear
F C
And there is no one there to dry your tears
Cm G
I could hold you for a million years
A7 D7 G
To make you feel my love




(Play normal 'D' and 'A' before using D7 and A7)

Country – Chords and lyrics –

Use Keyboard phone App

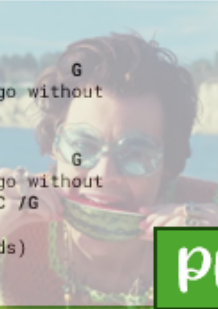
G Em D
Almost Heaven, West Virginia,
C G
Blue Ridge Mountains, Shenandoah River.
G Em
Life is old there, older than the trees,
D
younger than the mountains,
C G
growin' like a breeze.
[Chorus] G D
Country Roads, take me home,
Em C
to the place I belong
G D
West Virginia, mountain mama,
C G
take me home, country roads.



Modern pop – Chords and lyrics –

Use Keyboard phone App

Dm7 Am7
Tastes like strawberries On a summer evening
C G
And it sounds just like a song
Dm7 Am7
I want more berries and that summer feeling
C G
It's so wonderful and warm
Dm7 Am7
Breathe me in, breathe me out
C G
I don't know if I could ever go without
Dm7 Am7
I'm just thinking out loud
C G
I don't know if I could ever go without
[Chorus] Dm7 / Am7 / C / G
Watermelon sugar high
(4 times – cycle through chords)



Keywords:

Extended chords – Chords with notes beyond a basic triad.

Seventh Chords – Chords with a 7th note – adds drama.

Sus Chords – Chords with a suspended note (4th) – an unresolved sound.


Verse – The "Story" part of song, with words that rhyme but don't repeat.

Chorus – Main message of song – Words are the same each time.

Rock'n'Roll – Chords and lyrics

Use Keyboard phone app

D Dsus4 D Dsus4 D G
This thing, called love,
C G
I just can't handle it,
D Dsus4 D Dsus4
This thing, called love,
D G C G
I must, get 'round to it,
D
I ain't ready,
Bb C D
Crazy little thing called love.
D Dsus4 D Dsus4
This thing, called love,
D G C G
it cries, in a cradle all night,
D Dsus4 D Dsus4
It swings, it jives,
D G C G
it shakes all over like a jellyfish,
D
I kinda like it,
C D
Crazy little thing called love.



Playing / Analysing songs

Year 8

Physical Education

Term 5

1 Introduction to the Muscular System

Muscles enable movement by contracting and relaxing. There are over 600 muscles in the body, mainly skeletal muscles.

Every sport movement—from running to throwing—requires muscle use.



Muscular System

2 Major Muscles of the Body

Key muscles: biceps, triceps, deltoids, quadriceps, hamstrings, gluteals, etc. Each has a specific movement function.

e.g. Quadriceps extend the knee when kicking a football; deltoids lift arms in netball.



Muscular System

3 How Muscles Work in Pairs

Muscles work in **antagonistic pairs**: one contracts, the other relaxes. Agonist = working muscle, antagonist = opposing.

e.g. Sprinting uses hamstrings (agonist) and quadriceps (antagonist) in rapid alternation.



Muscular System

4 Types of Muscle Contractions

Types: **Concentric** (shortens), **Eccentric** (lengthens), **Isometric** (holds). Each helps control movement.

e.g. Wall sit = isometric, bicep curl = concentric/eccentric phases.



Muscular System

5 Introduction to the Cardiovascular System

Made up of the **heart, blood, and blood vessels**. Transports oxygen and nutrients.

Delivers oxygen to working muscles during games and training.

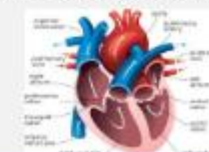


Cardiovascular System

6 Structure of the Heart

The heart has four chambers and works as a pump. Blood moves from the heart to lungs (to pick up oxygen) then around the body.

e.g. The heart works faster during high-intensity sport to pump more oxygen.



Cardiovascular System



How do plants use glucose?

Respiration

Production of large molecules:

- Starch
- Proteins
- Oils and fats

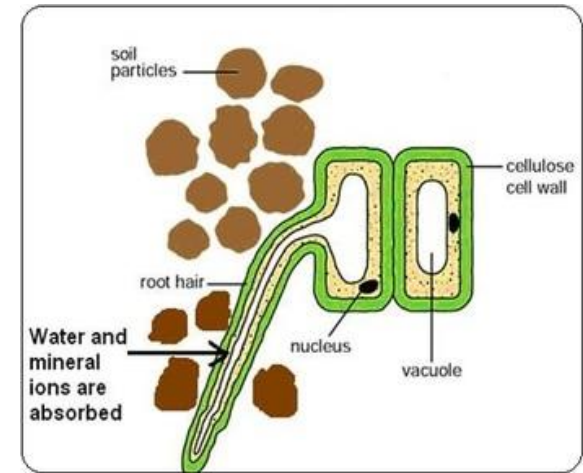
Roles of roots

- Stability of plant
- Absorb water
- Absorb minerals

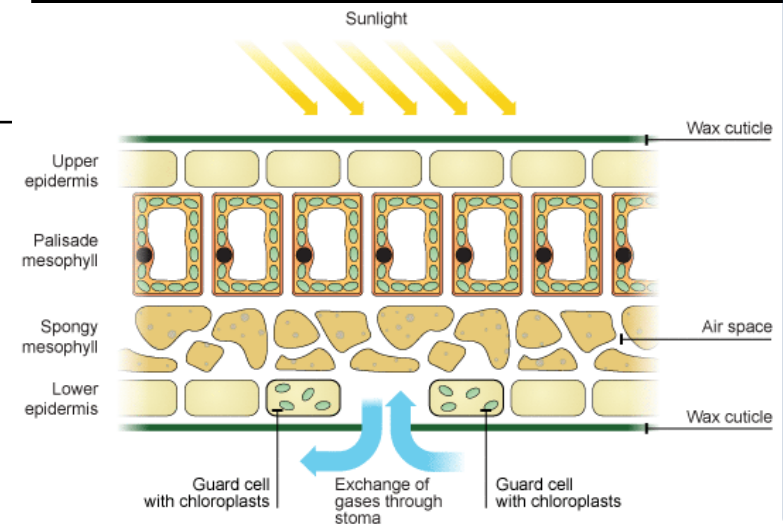
Factors affecting the rate of photosynthesis:

- Light intensity
- Colour of light
- Carbon dioxide concentration
- Availability of water
- Temperature
- Number and size of leaves

Root structure showing root hair cell



Cross section of a leaf:



Keyword	Definition
Photosynthesis	The chemical reaction by which plants make glucose using light, carbon dioxide and water
Xylem	Tissue that carries water from the roots to the leaves
Phloem	Tissue that carries dissolved sugars around the plant
Transpiration	The loss of water vapour through pores in the leaf
Transpiration stream	Movement of water through roots and xylem to the leaves
Reactants	Chemicals required for a chemical reaction
Product	Chemicals produced in a chemical reaction

Part of Plant	Adaptations
Roots	Root hair cells to increase surface area for absorption of water
Stem	Tube like xylem to carry water Tube like phloem to carry dissolved sugars
Leaf - Epidermis/cuticle	Reduces water loss and prevents the entry of pathogens
Leaf - Palisade mesophyll	Regular shaped cells with large numbers of chloroplasts to increase the rate of photosynthesis
Leaf - Spongy mesophyll	Provides air gaps to allow gas exchange (oxygen out, carbon dioxide in)
Leaf - Guard cells	Allow pores (stoma) to be opened and closed to control water loss

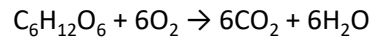
Keywords

Breathing	The inflation and deflation of the lungs by the contraction of the diaphragm and intercostal muscles.
Lung volume	Measure of the amount of air breathed in or out.
Respiration	Reaction which release energy from glucose.
Gas exchange	Process involving gases moving in and out of our blood.
Aerobic respiration	Breaking down glucose with oxygen to release energy and producing carbon dioxide and water.
Anaerobic respiration	Releasing energy from the breakdown of glucose without oxygen, producing lactic acid (in animals).
Fermentation	Yeast anaerobically respiring to produce ethanol and carbon dioxide.

Respiration

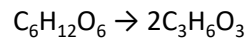
Aerobic respiration in animals:

Glucose + oxygen → carbon dioxide + water



Anaerobic respiration in animals:

Glucose → lactic acid



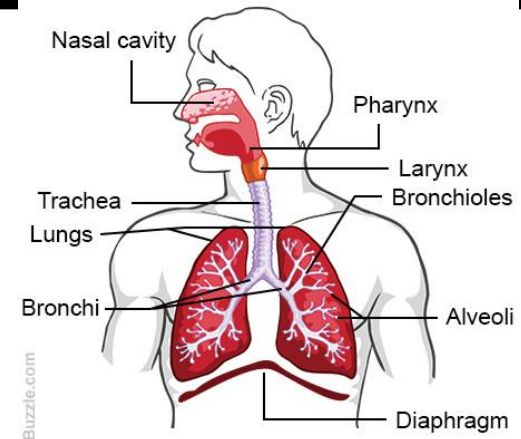
A poisonous waste product called lactic acid is produced, which can be removed by reacting it with oxygen:

Lactic acid + oxygen → carbon dioxide + water

Anaerobic respiration in yeast (fermentation):

Glucose → Ethanol + carbon dioxide

Structure of the Lungs

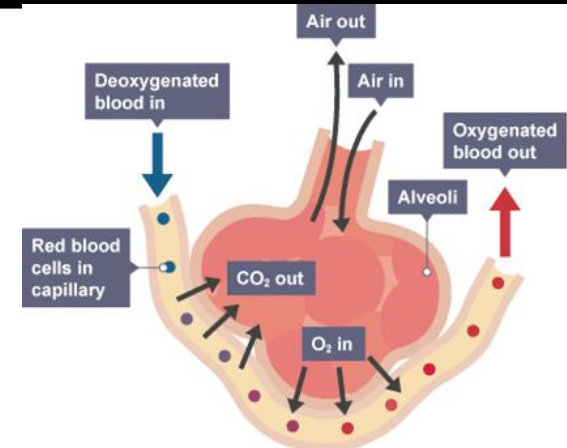


Trachea - Carries air from the mouth and nose to the lungs.
 Bronchi - Two tubes which carry air to the lungs.
 Bronchioles - Small tubes in the lung.
 Alveoli - Small air sacs found at the end of each bronchiole.
 Diaphragm - A sheet of muscle found underneath the lungs.

Gas Exchange

Gas exchange takes place in the alveoli, the tiny air sacs at the end of the bronchioles.

1. Oxygen molecules diffuse **from** the alveolus **into** the blood in the capillary, moving from higher to lower concentration.
2. Carbon dioxide molecules diffuse **from** the blood in the capillary **into** the alveolus, moving from higher to lower concentration.



Physical Change

Changes the physical state of a substance

No new substances are formed

Reversible - easily undone

No new matter created

Examples: change in state, making a mixture.

Chemical Change

New products are formed from the reactants

New substances are formed

Often irreversible.

No new matter created

Examples : Colour change, gas or heat given off.



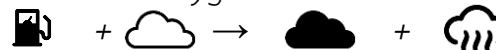
Fuel: Store of chemical energy which can be released as heat.

Combustion: A reaction with oxygen where energy is transferred to the surroundings as heat and light.

Types of combustion:

Complete combustion - plenty of oxygen

Fuel + oxygen → carbon dioxide + water



Incomplete combustion - not enough oxygen

Fuel + oxygen → carbon monoxide + carbon + water



Chemical Equations

Chemical equations tell us what happens in a chemical reaction

General: Reactant(s) → Product(s)

Example:
Word equation: Magnesium + oxygen → magnesium oxide



The law of conservation of mass

Atoms are not created or destroyed during reactions so:



Mass of reactants = mass of products



Thermal decomposition : A single reactant is broken down into simpler products by heating.

Example: *Metal carbonate → metal oxide + carbon dioxide*

Types of chemical reaction

+ O₂ **Oxidation:** The addition of oxygen
- Example: combustion

- O₂ **Reduction:** The removal of oxygen



Decomposition: Breaking down a compound
- Example: Thermal decomposition



Displacement: A more reactive substance takes the place of a less reactive substance



Neutralisation: Involves acids and bases