

KNOWLEDGE ORGANISER

YEAR 9 – TERM 4



Think Like An
Environmentalist

Community, Collaboration and Challenge

Long date and learning goal (LG) on left and underlined.

Minimise errors

Use pencil or blue or black pen for all written work, ruler for straight lines.

Worksheets glued in straight, next to margin.

Thursday 7th September 2023

LG: To identify and use terms that indicate the chronology of events.

<p>1 An autobiography is a story written in first chronological order containing your facts and opinions.</p>	<p>2 AS is a young child. I helped dad remove spiders from the garden shed. Some came later. I studied biology. Having reached adulthood I studied Central America. Later in life I presented the programme Animal legs and wings.</p>
--	---

I was born on May 19th 1995. At the age of 26 I signed a contract with Arsenal on 28th November 2021. A few years later I was part of the biggest football team that won the Euro 2022 tournament.

Monday 27th September 2023

LG: To identify and use terms that indicate the chronology of events.

MUSIC NOTES:

- Day: Monday
- Date: 27/9
- Time: 10:00
- Location: School
- Topic: Music
- Activity: Listening to music
- Notes: ...

1. The music was very good and we all enjoyed it.
2. The music was very good and we all enjoyed it.
3. The music was very good and we all enjoyed it.
4. The music was very good and we all enjoyed it.
5. The music was very good and we all enjoyed it.
6. The music was very good and we all enjoyed it.
7. The music was very good and we all enjoyed it.

Numbered questions written in margin.

Always try your best to be neat (even when note taking).

Feel proud

Single line through mistake. No scribbles.

Prepare for the future

Respond to feedback in green pen.

Easier to read and give better feedback

ATTENDANCE MATTERS



EVERY DAY COUNTS

Missing just 1 day every 2 weeks is the same as missing 10% of the school year.

LEARNING

Being in school allows you the best opportunity to learn.



WELLBEING

Attending school supports your mental and emotional health.

FUTURE SUCCESS

Regular attendance at school is vital for building the key skills needed for future employment



EQUIPMENT



School Bag



Knowledge Organiser



Black and Green Pens



Pencil case



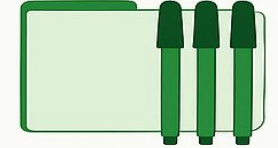
Calculator



Pencil



Rubber



Whiteboard and whiteboard pen



Highlighters



Ruler

SCHOOL DAY

9:00–9:05

AM Reg

9:05–10:20

Lesson 1

10:20–11:35

Lesson 2

11:35–12:05

Break 1

12:05–13:20

Lesson 3

13:20–13:50

Break 2

13:50–15:05

Lesson 4

15:05–15:30

PM Reg – assembly or guided reading

Multiplication Grid

x	1	2	3	4	5	6	7	8	9	10	11	12
1	1	2	3	4	5	6	7	8	9	10	11	12
2	2	4	6	8	10	12	14	16	18	20	22	24
3	3	6	9	12	15	18	21	24	27	30	33	36
4	4	8	12	16	20	24	28	32	36	40	44	48
5	5	10	15	20	25	30	35	40	45	50	55	60
6	6	12	18	24	30	36	42	48	54	60	66	72
7	7	14	21	28	35	42	49	56	63	70	77	84
8	8	16	24	32	40	48	56	64	72	80	88	96
9	9	18	27	36	45	54	63	72	81	90	99	108
10	10	20	30	40	50	60	70	80	90	100	110	120
11	11	22	33	44	55	66	77	88	99	110	121	132
12	12	24	36	48	60	72	84	96	108	120	132	144

PERIODIC TABLE OF ELEMENTS

Chemical Group Block



1																	18																																																							
1	1.0080															2	4.00260																																																							
1	H Hydrogen Nonmetal															2	He Helium Noble Gas																																																							
2	3	7.0	4	9.012183	Atomic Number										17	35.45	Atomic Mass, u																																																							
2	Li Lithium Alkali Metal	Be Beryllium Alkaline Earth Me...	Name										Cl Chlorine Halogen	Symbol																																																										
3	11	22.989...	12	24.305	Chemical Group Block										13	10.81	14	12.011	15	14.007	16	15.999	17	18.9984...	10	20.180																																														
3	Na Sodium Alkali Metal	Mg Magnesium Alkaline Earth Me...	3	4	5	6	7	8	9	10	11	12	Al Aluminum Post-Transition M...	Si Silicon Metalloid	P Phosphorus Nonmetal	S Sulfur Nonmetal	Cl Chlorine Halogen	Ar Argon Noble Gas	13	26.981...	14	28.085	15	30.973...	16	32.07	17	35.45	18	39.9																																										
4	19	39.0983	20	40.08	21	44.95591	22	47.867	23	50.9415	24	51.996	25	54.93804	26	55.84	27	58.93319	28	58.693	29	63.55	30	65.4	31	69.723	32	72.63	33	74.92159	34	78.97	35	79.90	36	83.80																																				
4	K Potassium Alkali Metal	Ca Calcium Alkaline Earth Me...	Sc Scandium Transition Metal	Ti Titanium Transition Metal	V Vanadium Transition Metal	Cr Chromium Transition Metal	Mn Manganese Transition Metal	Fe Iron Transition Metal	Co Cobalt Transition Metal	Ni Nickel Transition Metal	Cu Copper Transition Metal	Zn Zinc Transition Metal	Ga Gallium Post-Transition M...	Ge Germanium Metalloid	As Arsenic Metalloid	Se Selenium Nonmetal	Br Bromine Halogen	Kr Krypton Noble Gas	4	19	39.0983	20	40.08	21	44.95591	22	47.867	23	50.9415	24	51.996	25	54.93804	26	55.84	27	58.93319	28	58.693	29	63.55	30	65.4	31	69.723	32	72.63	33	74.92159	34	78.97	35	79.90	36	83.80																	
5	37	85.468	38	87.62	39	88.90584	40	91.22	41	92.90637	42	95.95	43	96.90636	44	101.1	45	102.9055	46	106.42	47	107.868	48	112.41	49	114.818	50	118.71	51	121.760	52	127.6	53	126.9045	54	131.29																																				
5	Rb Rubidium Alkali Metal	Sr Strontium Alkaline Earth Me...	Y Yttrium Transition Metal	Zr Zirconium Transition Metal	Nb Niobium Transition Metal	Mo Molybdenum Transition Metal	Tc Technetium Transition Metal	Ru Ruthenium Transition Metal	Rh Rhodium Transition Metal	Pd Palladium Transition Metal	Ag Silver Transition Metal	Cd Cadmium Transition Metal	In Indium Post-Transition M...	Sn Tin Post-Transition M...	Sb Antimony Metalloid	Te Tellurium Metalloid	I Iodine Halogen	Xe Xenon Noble Gas	5	37	85.468	38	87.62	39	88.90584	40	91.22	41	92.90637	42	95.95	43	96.90636	44	101.1	45	102.9055	46	106.42	47	107.868	48	112.41	49	114.818	50	118.71	51	121.760	52	127.6	53	126.9045	54	131.29																	
6	55	132.90...	56	137.33											72	178.49	73	180.9479	74	183.84	75	186.207	76	190.2	77	192.22	78	195.08	79	196.96...	80	200.59	81	204.383	82	207	83	208.98...	84	208.98...	85	209.98...	86	222.01...																												
6	Cs Cesium Alkali Metal	Ba Barium Alkaline Earth Me...											Hf Hafnium Transition Metal	Ta Tantalum Transition Metal	W Tungsten Transition Metal	Re Rhenium Transition Metal	Os Osmium Transition Metal	Ir Iridium Transition Metal	Pt Platinum Transition Metal	Au Gold Transition Metal	Hg Mercury Transition Metal	Tl Thallium Post-Transition M...	Pb Lead Post-Transition M...	Bi Bismuth Post-Transition M...	Po Polonium Metalloid	At Astatine Halogen	Rn Radon Noble Gas	6	55	132.90...	56	137.33											72	178.49	73	180.9479	74	183.84	75	186.207	76	190.2	77	192.22	78	195.08	79	196.96...	80	200.59	81	204.383	82	207	83	208.98...	84	208.98...	85	209.98...	86	222.01...
7	87	223.01...	88	226.02...											104	267.1...	105	268.1...	106	269.1...	107	270.1...	108	269.1...	109	277.1...	110	282.1...	111	282.1...	112	286.1...	113	286.1...	114	290.1...	115	290.1...	116	293.2...	117	294.2...	118	295.2...																												
7	Fr Francium Alkali Metal	Ra Radium Alkaline Earth Me...											Rf Rutherfordium Transition Metal	Db Dubnium Transition Metal	Sg Seaborgium Transition Metal	Bh Bohrium Transition Metal	Hs Hassium Transition Metal	Mt Meitnerium Transition Metal	Ds Darmstadtium Transition Metal	Rg Roentgenium Transition Metal	Cn Copernicium Transition Metal	Nh Nihonium Post-Transition M...	Fl Flerovium Post-Transition M...	Mc Moscovium Post-Transition M...	Lv Livermorium Post-Transition M...	Ts Tennessine Halogen	Og Oganesson Noble Gas	7	87	223.01...	88	226.02...											104	267.1...	105	268.1...	106	269.1...	107	270.1...	108	269.1...	109	277.1...	110	282.1...	111	282.1...	112	286.1...	113	286.1...	114	290.1...	115	290.1...	116	293.2...	117	294.2...	118	295.2...
			57	138.9055	58	140.116	59	140.90...	60	144.24	61	144.91...	62	150.4	63	151.964	64	157.2	65	158.92...	66	162.500	67	164.93...	68	167.26	69	168.93...	70	173.05	71	174.9668																																								
			La Lanthanum Lanthanide	Ce Cerium Lanthanide	Pr Praseodymium Lanthanide	Nd Neodymium Lanthanide	Pm Promethium Lanthanide	Sm Samarium Lanthanide	Eu Europium Lanthanide	Gd Gadolinium Lanthanide	Tb Terbium Lanthanide	Dy Dysprosium Lanthanide	Ho Holmium Lanthanide	Er Erbium Lanthanide	Tm Thulium Lanthanide	Yb Ytterbium Lanthanide	Lu Lutetium Lanthanide																																																							
			89	227.02...	90	232.038	91	231.03...	92	238.0289	93	237.04...	94	244.06...	95	243.06...	96	247.07...	97	247.07...	98	251.07...	99	252.0830	100	257.0...	101	258.0...	102	259.1...	103	266.1...																																								
			Ac Actinium Actinide	Th Thorium Actinide	Pa Protactinium Actinide	U Uranium Actinide	Np Neptunium Actinide	Pu Plutonium Actinide	Am Americium Actinide	Cm Curium Actinide	Bk Berkelium Actinide	Cf Californium Actinide	Es Einsteinium Actinide	Fm Fermium Actinide	Md Mendelevium Actinide	No Nobelium Actinide	Lr Lawrencium Actinide																																																							

01 Adjectives

THAT DESCRIBE:
age: young, old
colour: red, blue
condition: new, used
size: large, medium
speed: fast, slow
etc.

COMPARATIVE:
 smaller, better...

SUPERLATIVE:
 the smallest,
 the worst,
 the best...

08 Verbs

ACTION:
 to run, to organise,
 to read, to think...
 > Transitive
 or
 > Intransitive

LINKING:
 to be,
 to look, to appear,
 to seem, to smell...

**HELPING
 (= AUXILIARY):**
 can, may,
 will, must,
 should, to be,
 to have...

07 Pronouns

PERSONAL (subject):
 I, you, he, she, it,
 we,
 you, they

PERSONAL (object):
 me, you, him, her,
 it, us, you, them

PERSONAL (reflexive):
 myself, yourself,
 himself, herself,
 itself, ourselves,
 yourselves,
 themselves

DEMONSTRATIVE:
 this, these,
 that, those

POSSESSIVE:
 mine, yours, his,
 hers, its, ours,
 yours, theirs

INTERROGATIVE:
 how, where,
 when, which...?

INDEFINITE:
 somebody,
 anyone...

RELATIVE:
 that, which,
 whose, whom...

06 Prepositions

PLACE / DIRECTION:
 in, at, on,
 under, above,
 across,
 among,
 between...

TIME:
 in, at, on,
 over, until, about,
 during, before,
 after, while,
 through...

**OTHER (agent,
 phrase...):**
 by, with, on, over,
 to, up, within,
 beyond, for...

05 Nouns

COMMON NOUNS: house, dog, laptop...

PROPER NOUNS:
 (Capitalised)
 London, Paris,
 James, William,
 Julia, Jennifer...

> **VERBAL:** swimming...

> **COLLECTIVE:** choir, jury...

> **COMPOUND:** mother-in-law...

> **COUNTABLE:** book, day...

> **UNCOUNTABLE:** traffic, calm...

> **ABSTRACT V. CONCRETE:** wit vs. road...

02 Adverbs

PLACE:
 here, there,
 outside, everywhere,
 upstairs, nowhere,
 somewhere....

TIME:
 ago, before, since,
 yet, for, still,
 afterwards...

FREQUENCY:
 often, never,
 sometimes, always

MANNER:
 just, quite,
 quickly, hardly,
 well, carefully,
 barely, almost,
 scarcely,
 beautifully...

03 Conjunctions

COORDINATING:
 and, or, but,
 yet, nor, for, so

CORRELATIVE:
 both... and...,
 either... or...,
 just as... so...,
 whether... or...,
 neither... nor...,
 not only... but also...

SUBORDINATING:
 after, since, if,
 while, although,
 before, because,
 unless

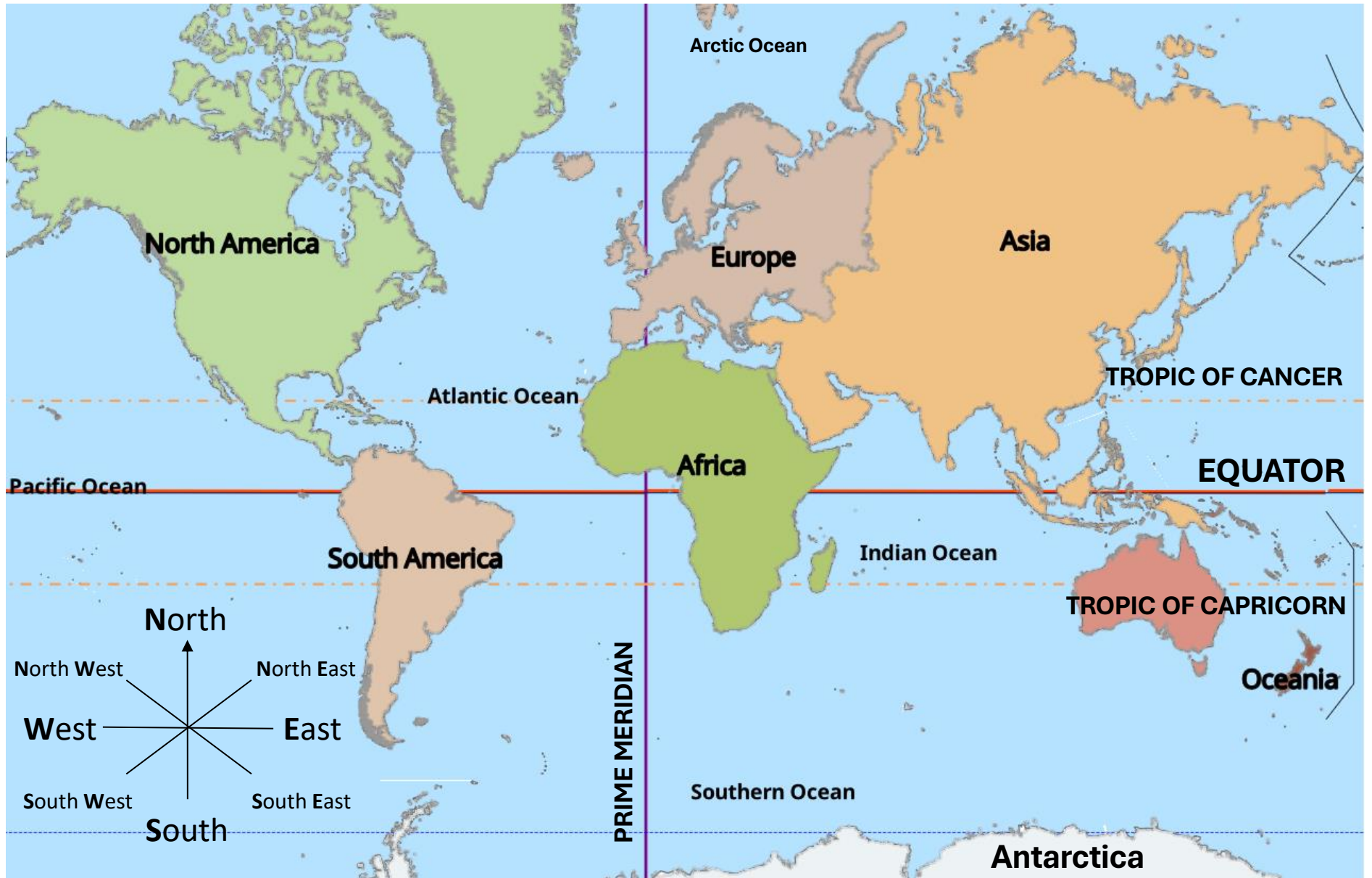
04 Determiners

TELLS US WHICH:
 each, every,
 some, none,
 all...

TELLS US WHOSE:
 my, your, her, his, its,
 our, your, their (= possessive
 adjectives or determiners)



World Map



Year 9 Hockney Portraits

Art

Term 4

Introduction to David Hockney

- **Objective:** Understand who David Hockney is and explore his contribution to modern portraiture.
- **Activities:**
 - Presentation on Hockney's life and key artworks.
 - Class discussion: What makes his portraits unique?
 - Sketchbook activity: Annotate and respond to one Hockney portrait.



Style Analysis

- **Objective:** Identify and analyze key features of Hockney's portrait style.
- **Activities:**
 - Examine composition, color, pattern, and perspective in Hockney's work.
 - Colour pencil/paint response studies in sketchbook.



Drawing Facial Features

- **Objective:** Practice drawing realistic facial features.
- **Activities:**
 - Guided drawing exercises (eyes, nose, mouth, ears).
 - Compare realistic drawing to Hockney's stylized approach.
 - Sketchbook work: Combine a realistic eye with Hockney-style background.
 - Practice drawing a full face from a reference image.

Compositional Planning

- **Objective:** Design a portrait composition inspired by Hockney's style.
- **Activities:**
 - Create thumbnail sketches exploring color, shape, and layout.
 - Experiment with simplified backgrounds and strong lines.
 - Annotate decisions in sketchbooks.
 - Finish chosen thumbnail in color.



Color Theory + Media Exploration

- **Objective:** Explore Hockney's use of color and media.
- **Activities:**
 - Media tests: watercolour, colored pencils, iPad drawing (if available).
 - Recreate a section of a Hockney portrait in chosen media.
 - **Homework:** Write a reflection on preferred media and why.



Begin Final Piece

- **Objective:** Start the final Hockney-inspired portrait.
- **Activities:**
 - Draw outline using reference photo and chosen style.
 - Begin applying color and line work.
 - Peer feedback sessions.
 - Final touches and class display.
 - Self and peer assessment using success criteria.



Computing

The Binary Number System

In the number system that us humans use, the number 10 means 'ten' because the digits mean '1 ten and 0 ones'.
 ...the number 33 means 'thirty three' because the digits mean '3 tens and 3 ones'.
 ...the number 528 means 'five hundred and twenty eight' because the digits mean '5 hundreds, 2 tens and 8 ones'.

10's	1's	
1	0	
10's	1's	
3	3	
100's	10's	1's
5	2	8

Humans developed this 'base 10' number system millions of years ago because they learnt to count using their hands, which have 10 digits. Computers, being electronic, are simply made up of switches (which can be in only one of two states (on/off)) and as a result cannot count like us! As a result, computers use a different number system – the binary number system

2's	1's		
1	0		
8's	4's	2's	1's
1	0	0	1

The columns in the binary system, from right to left are 1, 2, 4, doubling as we go ...in binary the digits 10 means 'the number two' because the digits mean '1 two and 0 ones'.
 ...the digits 1001 means 'the number nine' because the digits mean '1 eight, 0 fours, 0 twos and 1 one'.

Converting from Binary to Decimal/Denary

To convert a binary number into decimal/denary, the process is thankfully really easy! All we need to do is add up the column values which contain a one and ignore the column values which contain a zero.

For example, the following binary number has the decimal/denary value of 155. This is because the 1s in the binary number represents $128 + 16 + 8 + 2 + 1 = 155$

128	64	32	16	8	4	2	1
1	0	0	1	1	0	1	1

→ 155

Converting from Decimal/Denary to Binary

Converting from decimal/denary to binary, is also not too hard! We just need to work out which of the column values add together to form the decimal value that we needed to convert.

The easiest way is to do this work from left to right along the binary column values and if the column value can fit into our decimal number, we place a 1 under that value's column, subtract the column value from the decimal number and continue the process. For example, if we want to convert the decimal number 202 into binary, we would do the following:

Does 128 fit into 202? Yes! So, we add a 1 to the column and subtract 128 from 202 leaving 74

Does 64 fit into 74? Yes! So, we add a 1 to the column and subtract 64 from 74 leaving 10

Does 32 fit into 10? No! So, we add a 0 to the column and move on!

This process continues until we've made the remaining value of ten by placing 1s under the 8 and 2 columns (because $8 + 2 = 10$), and zeros under the others.

128	64	32	16	8	4	2	1
1	0	0	1	1	0	1	1

Binary Addition

Adding binary numbers is much like adding denary numbers (the system we use as humans).

Things to remember:

- Keep your numbers in the correct columns
- $1+1 = 10$ in binary
- $1+1+1 = 11$ in binary

We begin, like we would normally when adding two numbers together, with the right most column.

In this example, we add two 1s together, which of course is 2! However, in binary, 2 is represented as 10, and as there are two digits in this answer, we place the right digit (zero) under the column and carry the left digit (one) over to the next (left) column.

0 0 1 1 0 0 1 1	0 0 1 1 0 0 1 1
1 0 0 0 1 1 1 1	1 0 0 0 1 1 1 1
0	

Now we focus on the second column from the right. Here we have $1+1+1$ (including the carry), which of course equals 3. But in binary 3 is represented as 11. We therefore place the right digit underneath and carry the left digit.

0 0 1 1 0 0 1 1	0 0 1 1 0 0 1 1
1 0 0 0 1 1 1 1	1 0 0 0 1 1 1 1
0 0 1 0	0 0 1 0

This process then continues, moving through the columns to the left, until we have added the binary numbers.

0 0 1 1 0 0 1 1	0 0 1 1 0 0 1 1
1 0 0 0 1 1 1 1	1 0 0 0 1 1 1 1
0 0 0 1 0	0 0 1 1 0 0 1 1
1 0 0 0 1 1 1 1	1 0 0 0 1 1 1 1
1 1 0 0 0 0 1 0	1 0 0 0 1 1 1 1

Representing Characters

- As we know, computers can only deal with 0s and 1s (binary).
- All data that it needs to work with (numbers, sound, images etc) must be converted into binary for the computer to be able to process it.
- It is exactly the same for text, or one piece of text known as a character.
- Each time you hit a key on a keyboard, the computer generates a code for that letter, which is then processed by the CPU and the result might be the letter appearing on the screen or being printed on paper.
- So that all computer systems behave in a similar way it is important that there is an agreed set of codes for characters.
- The agreed set of codes to represent the main characters in the English language is known as ASCII (American Standard Code for Information Interchange).

Below, you can see that each character is represented by a number. The binary table contains 5 binary numbers, and by working out the value of each binary number, we can see which letter it represents by looking it up in the ASCII table.

ASCII TABLE

Decimal	Hex	Char	Decimal	Hex	Char	Decimal	Hex	Char
0	00	SPACE	16	10	P	32	20	R
1	01	START OF HEADING	17	11	Q	33	21	S
2	02	START OF TEXT	18	12	R	34	22	T
3	03	END OF TEXT	19	13	S	35	23	U
4	04	END OF LINE	20	14	T	36	24	V
5	05	START OF LINE	21	15	U	37	25	W
6	06	PRINTING RANGE	22	16	V	38	26	X
7	07	DELETE	23	17	W	39	27	Y
8	08	BACKSPACE	24	18	X	40	28	Z
9	09	SHIFT	25	19	Y	41	29	[
10	0A	SHIFT	26	1A	Z	42	2A]
11	0B	SHIFT	27	1B	[43	2B	^
12	0C	SHIFT	28	1C]	44	2C	_
13	0D	SHIFT	29	1D	^	45	2D	~
14	0E	SHIFT	30	1E	_	46	2E	
15	0F	SHIFT	31	1F		47	2F	
16	10		32	20		48	30	
17	11		33	21		49	31	
18	12		34	22		50	32	
19	13		35	23		51	33	
20	14		36	24		52	34	
21	15		37	25		53	35	
22	16		38	26		54	36	
23	17		39	27		55	37	
24	18		40	28		56	38	
25	19		41	29		57	39	
26	1A		42	2A		58	3A	
27	1B		43	2B		59	3B	
28	1C		44	2C		60	3C	
29	1D		45	2D		61	3D	
30	1E		46	2E		62	3E	
31	1F		47	2F		63	3F	
32	20		48	30		64	40	
33	21		49	31		65	41	
34	22		50	32		66	42	
35	23		51	33		67	43	
36	24		52	34		68	44	
37	25		53	35		69	45	
38	26		54	36		70	46	
39	27		55	37		71	47	
40	28		56	38		72	48	
41	29		57	39		73	49	
42	2A		58	3A		74	4A	
43	2B		59	3B		75	4B	
44	2C		60	3C		76	4C	
45	2D		61	3D		77	4D	
46	2E		62	3E		78	4E	
47	2F		63	3F		79	4F	
48	30		64	40		80	50	
49	31		65	41		81	51	
50	32		66	42		82	52	
51	33		67	43		83	53	
52	34		68	44		84	54	
53	35		69	45		85	55	
54	36		70	46		86	56	
55	37		71	47		87	57	
56	38		72	48		88	58	
57	39		73	49		89	59	
58	3A		74	4A		90	5A	
59	3B		75	4B		91	5B	
60	3C		76	4C		92	5C	
61	3D		77	4D		93	5D	
62	3E		78	4E		94	5E	
63	3F		79	4F		95	5F	

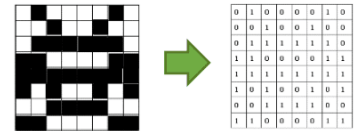
What is my name?

64	32	16	8	4	2	1	
1	0	1	0	0	1	0	82 - R
0	1	0	0	1	0	1	73 - I
1	0	0	1	1	1	0	78 - N
1	0	0	0	1	1	1	71 - G
1	0	0	1	1	1	1	79 - O

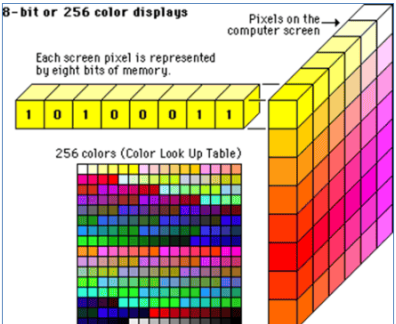
Representing Images

Bitmaps

Bitmap images are made up of rows of "dots" called "pixels" (picture elements). Each pixel is represented by a binary number. Behind the scenes, this 1-bit image (with each shade represented by a bit) is in fact a series of numbers:



In a coloured bitmap, longer binary numbers represent a different colour:



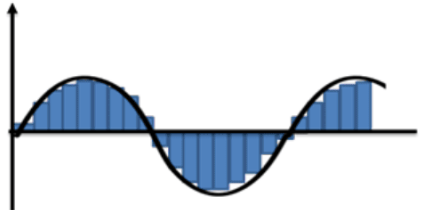
As images get more colourful, longer binary numbers are needed so that a bigger combination of colours can be shown.

Representing Sound

Analogue sounds (sound waves that continuously vary) are pure and of perfect quality. However, computer recorded sound is not pure, not real and not of perfect quality and this is because sound has been digitised – it has been sampled at set intervals.

Sampling

Sampling is the process by which computers digitise sound. They measure the height of sound waves at regular intervals and record the measurement as a binary number. So, whereas analogue sound is continuous over time, digitised sound is made up of lots of 'sound bites' over time.



When computers play sound through a speaker, they process each of the binary measurements and send signals to the speaker making it vibrate in different ways, according to the binary data.

Key Vocabulary

Key Word	Definition
Denary / Decimal Binary	Base 10 number system - The number system we learnt in primary school Base two number system – the only number system computers know – they can only understand two digits because they are made up of switches that can only be in the on (1) and off (0) state.
ASCII	Universally accepted binary numbers for each keyboard character
Bitmap	A computer image file which is made up of tiny pixels of colour. Each pixel is represented by a set of binary bits and mapped to the screen
Sampling	Recording analogue sound at regular intervals and converting each snippet of sound to a binary value.
Digital Processing	Applying maths on the binary which represents sound in order to manipulate how it sounds.

Year 9

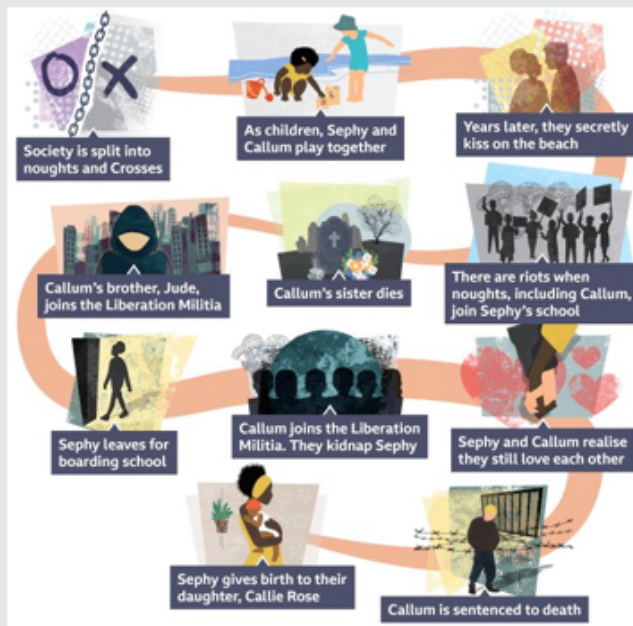
Drama

Term 4

Plot

Noughts & Crosses tells the story of Sephy and Callum, two young people from different races who fall in love and face prejudice, terrorism, and injustice.

The play takes place in a world similar to our own, but with a large divide between the ruling class, the Crosses, and the underclass, the noughts. The Crosses are dark-skinned and privileged, while the noughts are light-skinned and poor.



Terminology

Proxemics - the deliberate use of space, distance, and positioning between actors on stage to communicate unspoken information about their relationships, status, emotions, and the situation to the audience.

Semiotics - the study of how signs and symbols create meaning in theatre and performance.

Conscience Alley - a character facing a dilemma walks down an "alley" formed by two lines of people, who present opposing viewpoints (e.g., "do it" vs. "don't do it") to influence their decision.

Characters

Sephy - a privileged, dark-skinned Cross.

Callum - a poor, light-skinned Nought.

Kamal - a powerful, wealthy politician and Sephy's father.

Jasmine - Sephy's mother

Minerva - Sephy's sister

Meggie - Callum's mother

Ryan - Callum's father

Jude - Callum's brother

Lynette - Callum's sister

Mr Corsa - Headmaster at Heathcroft School

Shania - a Nought girl who attends Heathcroft School

English



'Hitcher,' by Simon Armitage

The poem is narrated by a speaker who describes picking up a hitchhiker and then violently attacking him. The speaker is under pressure at work and feels disconnected from his own life. The hitchhiker, on the other hand, represents freedom and a carefree existence, which the speaker envies and resents.

Themes

1. Violence and Aggression: The poem vividly depicts an act of unprovoked violence, highlighting the speaker's pent-up anger and frustration. 2. Freedom vs. Conformity: The hitchhiker symbolizes a free-spirited lifestyle, in stark contrast to the speaker's mundane and pressured existence. 3. Alienation and Disconnection: The speaker feels alienated from his own life and disconnected from societal norms, which drives his violent behavior.

'Half-Caste,' by John Agard

Dramatic and sarcastic, this monologue poem attacks the derogatory term for mixed-race individuals. Using Caribbean Creole, the poet challenges the notion of being 'half' a person, using analogies like Picasso's art and Tchaikovsky's symphony to argue that blending is a complex, artistic, and whole experience.

Wilfred Owen's 'Anthem for Doomed Youth' is a powerful WWI poem mourning young soldiers lost in battle, ironically contrasting the harsh reality of trench warfare with traditional, inadequate funeral rites, suggesting that the sounds of guns replace church bells and the pallor of grieving loved ones serves as the only "pall" for fragmented bodies, highlighting the brutal, dehumanizing nature of war. The poem serves as a bitter commentary on the disconnect between official ceremonies and the brutal experience of death on the front lines, lamenting the silenced lives and the hollow memorials.

'Havisham,' by Carol Ann Duffy

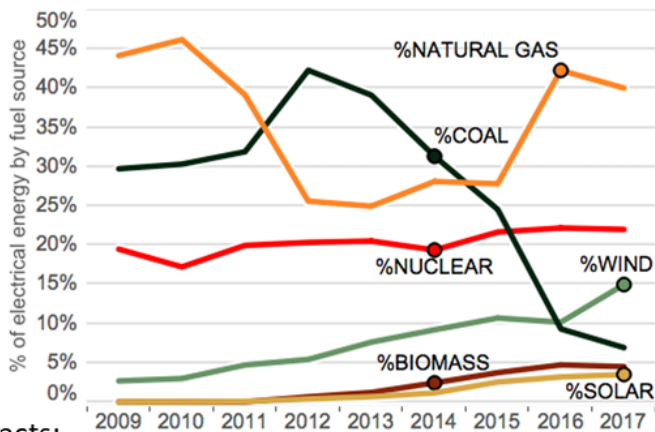
This poem is a dramatic monologue exploring the corrosive nature of love turning to hate through the voice of Charles Dickens' *Great Expectations* character, Miss Havisham. The speaker, a jilted bride, expresses intense, violent rage and enduring bitterness while sitting in her decaying wedding dress.

- **Perspective:** Written in the first person, giving voice to a scorned woman.
- **Themes:** Explores themes of heartbreak, misogyny, insanity, and the destructive, all-consuming nature of revenge.
- **Imagery:** Uses morbid imagery to show the merging of love and hate, such as "beloved sweetheart bastard" and "dark green pebbles for eyes".
- **Structure:** The poem breaks down logical thought, reflecting the character's fragmented, chaotic mental state, and obsession with the past.

UK Energy Trend

The UK consumes less energy than it did in 1970, despite the population being 9.1 million higher. 12% less energy is used by the average household. Heavy industry uses 60% less energy due to its decline. Demand for energy by transport has increased.

UK's Energy Mix



Key facts:

- 2015 – majority of UK's energy mix = fossil fuels
- Nuclear provides just over 1/5th of the UK's energy mix
- Renewables provide just over 20% of the UK's energy mix
- 2011 coal use increased - older power stations worked to capacity (due to closure - EU regulations on emissions)
- Oil and gas reserves have declined.
- Renewable energies like wind are growing in significance, but are still only a small percentage of energy produced
- Renewables are encouraged to meet targets on reducing emissions

Energy Exploitation Issues Economic vs Environmental

Fossil fuels

- Unustainable, they will eventually become too expensive or run out.
- Costs increase to deal with the effects of climate change and adaption to it.
- CO₂ is released which contributes to acid rain and climate change.
- Fracking for shale gas can cause earthquakes and groundwater pollution.

Nuclear

- Nuclear plants are expensive to build and decommission.
- Transporting and storing nuclear waste is expensive.
- Waste is radioactive for 100 years+ and has to be stored safely to avoid contamination.
- Nuclear accidents have long-term impacts on people and wildlife.

Renewable energy resources

- High set-up costs. Costs increase in remote areas.
- Biomass can reduce land available for food production increasing food prices.
- Low profitability is a concern.
- Biomass reduces biodiversity as only one crop is grown e.g. sugar cane.
- HEP schemes flood land upstream, changing the landscape and wildlife.
- Wind turbines can affect bird migration.

Option C: Energy Tier 3

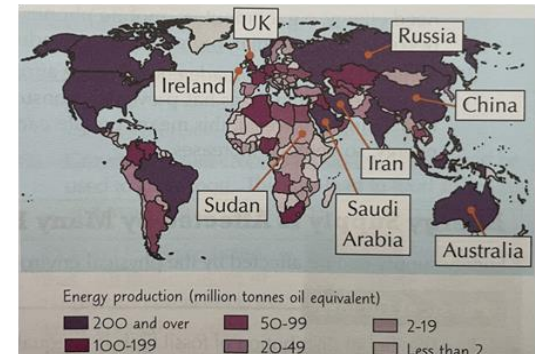
Energy Security – having a reliable, uninterrupted & affordable supply of energy.

Energy Surplus – where a country produces more energy than it's population requires.

Energy Deficit – where a country has less energy than it requires

Energy Insecurity – where a country does not have a reliable & affordable supply of energy, or it is interrupted

Global Energy Production

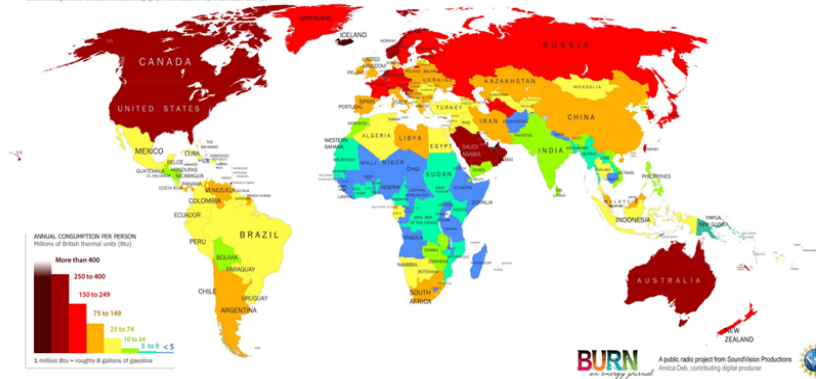


Total energy produced per country, 2012
Production is unevenly distributed:

Some countries produce lots of energy due to large energy reserves & money for exploitation (Russia – oil & gas)
Some countries have little energy - few resources or unable to exploit resources due to poverty/ war (Ireland – few resources, Sudan – war & poverty)

⚡ Global Energy Consumption

Energy Consumption Per Person, by country, 2010.



Wealth links to Energy Consumption:

Wealthy = higher consumption due to higher standard of living & because they can afford it – most use energy-intensive devices – e.g. Sweden, USA.

Poorer, less developed = lower consumption as people are less able to afford it, less energy is available & less dependency on energy consumption – e.g. Mongolia.

⚡ Reasons for Rising Global Energy Demand

1. Rising Population

- Global population to increase to over 9 billion by 2040
- More people means more energy needed e.g. to heat homes

2. Economic Development

- Recent economic development has increased LIC wealth:
- People use wealth to buy more material possessions that use more energy e.g. cars, TVs
 - As countries develop, industry expands which uses more energy

3. Advances in Technology

- Advances have developed new energy powered devices
- Some advances have made energy more affordable – Wind energy e.g. in Texas has limited consumer energy costs

⚡ Impacts of energy insecurity

Environmental/Economic Costs:

- Fossil fuels are used up, less accessible reserves means more environmentally sensitive areas are exploited
- Higher costs & increased risk of environmental damage

Food Production:

- Insecurity can limit use of agricultural machinery
- Demand for cleaner & cheaper energy – increases biofuel demand
- Growing biofuel

Industrial Output:

- Energy shortages & higher energy costs reduced industrial output – factories produce less or relocate to minimise costs causing job insecurity
- Higher costs often passed to the consumer

Potential for Conflict:

- Potential for political instability when demand exceeds supply
- Causes conflict between countries with an energy surplus & those with a deficit e.g. South Sudan vs Sudan conflict due to ownership of Heglig oil field resulted in a short war

⚡ Increasing Energy Supply



Biomass
Energy produced from organic matter



Wind
Turbines on land or at sea generate electricity



Hydro (HEP)
Largescale dams or micro dams turn turbines



Tidal
Turbines within barrages across river estuaries



Geothermal
Water heated underground in contact with hot rocks produces steam



Wave
Waves force air into a chamber which turns a turbine



Solar
Photovoltaic cells convert sunlight into electricity

⚡ Factors Affecting Global Energy Supply

Physical Factors:

- Global distribution is uneven, therefore geographically some countries have energy, some do not
- Location of fossil fuel reserves affects cost and ease of extraction
- Areas climate of geography determines potential to generate renewable energy e.g. solar or wind power

Technological Factors:

- Advances in technology means its more possible to exploit new resources (fracking to reach gas)
- Some technology makes it easier to exploit existing resources e.g. autonomous machines to extract in risky or hard to reach areas
- Technology can safeguard transport of existing resources e.g. earthquake proof gas pipelines from Alaska to the USA

Economic Factors:

- Remaining non-renewable resources costly to extract
- LICs may have potential energy resources but are too poor to exploit them
- Building new energy infrastructure (nuclear power stations, wind farms) can be expensive – e.g. Hinkley Point C £23 billion

Political Factors:

- Political instability in countries with large reserves can limit ability to export – Middle East instability has encouraged countries to seek oil elsewhere
- Climate Change has led to international agreements to reduce concentration of greenhouse gases e.g. Paris Agreement – countries pledged to reduce CO2
- German government plans to phase out nuclear power stations by 2022

📖 Natural Gas Tier 3 Vocab

Fossil fuel - A natural fuel formed in the geological past from the remains of living organisms, such as coal or natural gas.

Shale gas - A form of natural gas (mostly methane), found underground in shale rock.

Carbon emissions - Carbon compound (such as carbon dioxide) released into the atmosphere, often through human activity such as the burning of fossil fuels such as coal or gas.

👍 Advantages

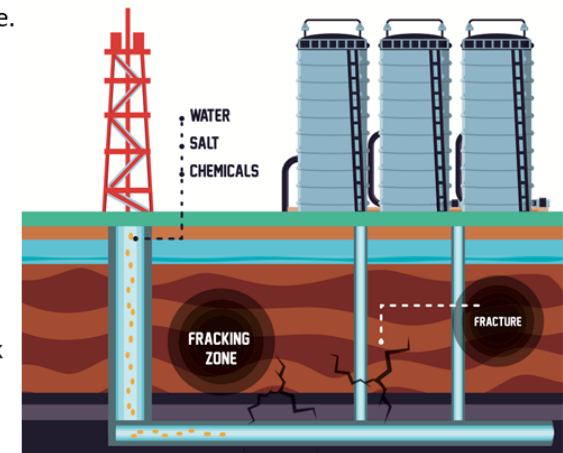
- 👤 Natural gas is one of the cleanest fossil fuels, releasing 45% fewer carbon emissions than coal and oil.
- 👤 It does not produce waste products, such as coal ash.
- 👤 If there is a gas leak, it disappears quickly into the air, whereas oil leaks have a devastating impact on the environment and are expensive to clean.
- 👤 Gas can be easily transported through pipelines directly to industry and residential properties.
- 👤 There are considerable reserves of gas around the world.
- 👤 Gas can be used for both heating and cooking.

👎 Disadvantages

- 👤 Gas is highly flammable, which means leaks can result in explosions.
- 👤 Natural gas is toxic.
- 👤 Gas infrastructure is expensive; pipelines cost a significant amount of money to construct.
- 👤 Unless an odorant is added to the gas, leaks can go undetected.
- 👤 The controversial process of fracking has caused earthquakes.
- 👤 Burning natural gas results in the release of greenhouse gases, particularly carbon dioxide.

Fracking

Recently, shale gas has been extracted using controversial fracking techniques. Fracking involves injecting bedrock with water and chemicals to extract gas.



Chambamontera Micro-Hydro Scheme

Hydro-electric power – Electricity generated by turbines that are driven by moving water.

Micro-hydro scheme – A small-scale hydroelectric scheme that generates between 5 kW and 100 kW of electricity.

Renewable energy source – A resource which is not diminished when it is used; it recurs and cannot be exhausted, such as wind and tidal energy.

Sustainable energy supply – Energy that can potentially be used well into the future without harming future generations.

Where is Chambamontera?

Chambamontera is a remote community in Peru in the Andes Mountains. Its closest town is Jaén.

Why was the scheme needed?

The community of Chambamontera required a sustainable energy scheme because most of its inhabitants relied on subsistence farming, small-scale coffee production, and livestock rearing.

The lack of electricity has significantly hindered development, as it is essential for heating, lighting, and powering homes and businesses.

Despite efficient farming practices, poverty was widespread, with nearly half of the population living on just US\$2 a day.

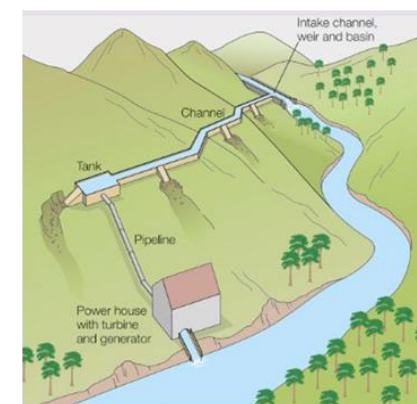
The area's challenging terrain, with steep slopes at altitudes up to 1700 meters and roads often closed in winter, contributes to its isolation and the economic impracticality of constructing a traditional electricity grid.

Main features

To address the energy shortage, Chambamontera implemented a micro-hydro scheme with the help of the charity Practical Action. The region's high rainfall, steep terrain, and fast-flowing rivers make it well-suited to harnessing water power as a sustainable energy source. The project's total cost was US\$51,000, with funding from the government, investments from Japan, and contributions from the local community. Each family faced an average cost of US\$750, with credit options available to assist with payments.

Benefits of the scheme

- The advantages of the scheme include:
- Business growth, as access to piped water has enabled the use of machinery for coffee processing.
 - Reduced risk of fires since there is no longer a need for kerosene lighting.
 - Improved educational opportunities by enabling students to study after dark due to improved school facilities.
 - Strengthened healthcare services through the refrigeration of medicines.
 - Promoting safer night-time mobility with the installation of streetlights.
 - A reduction in rural-to-urban migration.
 - Reduced flood risks by regulating river flow.
 - Decreased reliance on wood as a heating source helping to prevent deforestation and soil erosion.
 - A project lifespan of at least 25 years.
 - Consistent electricity during winter months when the demand for heating is greater.
 - A dependable power supply for refrigeration, lighting, computer use, and entertainment.



Deduction

@winsto_maths

What do I need to be able to do?

to do?

By the end of this unit you should be able to:

- Identify angles in parallel lines
- Solve angle problems
- Make conjectures with angles
- Make conjectures with shapes

Keywords

Parallel: two straight lines that never meet with the same gradient

Perpendicular: two straight lines that meet at 90°

Transversal: a line that crosses at least two other lines

Sum: the result of adding two or more numbers

Conjecture: a statement that might be true but is not proven

Equation: a statement that says two things are equal

Polygon: a 2D shape made from straight edges

Counterexample: an example that disproves a statement

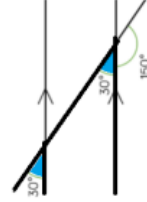
Alternate angles



Because alternate angles are equal the highlighted angles are the same size

R

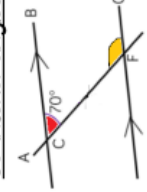
Corresponding angles



Because corresponding angles are equal the highlighted angles are the same size

R

Co-interior angles



Because co-interior angles have a sum of 180° the highlighted angle is 110°

R

As angles on a line add up to 180° co-interior angles can also be calculated from applying alternate/ corresponding rules first

Solving angle problems

Angles on a straight line 180°

Link angle facts to algebra



Form an equation

$$2x + 4x = 180^\circ$$

Solve

$$2x + 4x = 180^\circ$$

$$6x = 180^\circ$$

$$x = 30^\circ$$

State the reason

The sum of angles on a straight line is 180°

Vertically opposite angles

Equal

Angles around a point

360°

Triangles

Sum of angles is 180°

Isosceles have the same base angles



Interior Angles

The angles enclosed by the polygon



$$(\text{number of sides} - 2) \times 180$$

Making conjectures with angles



Proving a conjecture

A pattern is noticed for many cases



Apply the angle rules

The sum of angles in a triangle is 180°

Test the theory

$$180 - 70 - 20 = 90$$

$$180 - 85 - 5 = 90$$

$$180 - 45 - 45 = 90$$

Make a conjecture

The angle that meets the circumference in a semicircle is 90°

Disproving a conjecture

Only one counterexample is needed to disprove a conjecture

Making conjectures with shapes

Keywords and facts to recall with shape

Area: the amount of space inside a shape
Perimeter: the length around a shape
Regular Polygons: All sides and angles are equal

Quadrilateral Facts

Square

All sides equal size
All angles 90°
Opposite sides are parallel



Rectangle

Opposite sides are parallel



Rhombus

All sides equal size
Opposite angles are equal



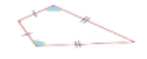
Parallelogram

Opposite sides are parallel
Opposite angles are equal
Co-interior angles



kite

No parallel lines
Equal lengths on top sides
Equal lengths on bottom sides
One pair of equal angles



Rotation & Translation

@whisto_maths

What do I need to be able to do?

- By the end of this unit you should be able to:
 - Identify the order of rotational symmetry
 - Rotate a shape about a point on the shape
 - Rotate a shape about a point not on a shape
 - Translate by a given vector
 - Compare rotations and reflections

Keywords

Rotate: a rotation is a circular movement

Symmetry: when two or more parts are identical after a transformation

Regular: a regular shape has angles and sides of equal lengths

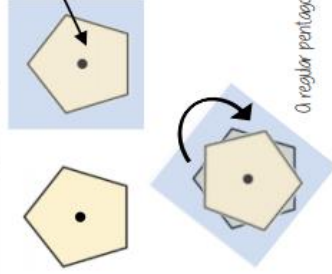
Invariant: a point that does not move after a transformation

Vertex: a point two edges meet

Horizontal: from side to side

Vertical: from up to down

Rotational Symmetry

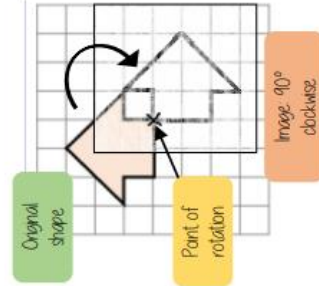


Tracing paper helps check rotational symmetry

- Trace your shape (mark the centre point)
- Rotate your tracing paper on top of the original through 360°
- Count the times it fits back into itself

A regular pentagon has rotational symmetry of order 5

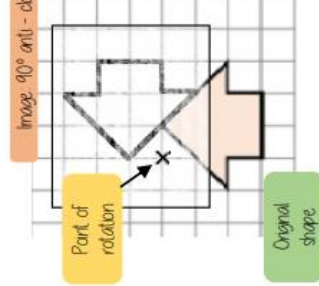
Rotate from a point (in a shape)



- Trace the original shape (mark the point of rotation)
- Keep the point in the same place and turn the tracing paper
- Draw the new shape



Rotate from a point (outside a shape)



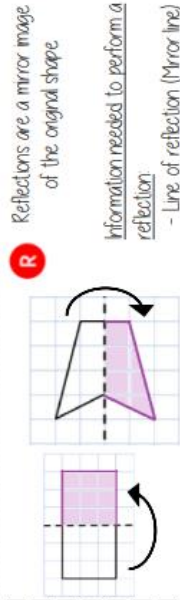
- Trace the original shape (mark the point of rotation)
- Keep the point in the same place and turn the tracing paper
- Draw the new shape

Translation and vector notation



Every vertex has been translated by the same amount

Compare rotations and reflections



Reflections are a mirror image of the original shape

Information needed to perform a reflection

- Line of reflection (Mirror line)

Rotations are the movement of a shape in a circular motion

Information needed to perform a rotation

- Point of rotation
- Direction of rotation
- Degrees of rotation

Pythagoras' theorem

What do I need to be able to do?

- By the end of this unit you should be able to:
- Use square and cube roots
 - Identify the hypotenuse
 - Calculate the hypotenuse
 - Find a missing side in a Right angled triangle
 - Use Pythagoras' theorem on axes
 - Explore proofs of Pythagoras' theorem

Keywords

- Square number:** the output of a number multiplied by itself
- Square root:** a value that can be multiplied by itself to give a square number
- Hypotenuse:** the largest side on a right angled triangle. Always opposite the right angle.
- Opposite:** the side opposite the angle of interest
- Adjacent:** the side next to the angle of interest

Squares and square roots **R**



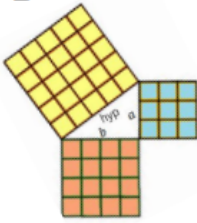
This can also be written as 6^2

e.g. $\sqrt{64} = 8$
Because $8 \times 8 = 64$

1 × 1	2 × 2	3 × 3	4 × 4	5 × 5	6 × 6	7 × 7	8 × 8	9 × 9	10 × 10
1	4	9	16	25	36	49	64	81	100

Square numbers

Determine if a triangle is right-angled

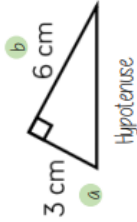


If a triangle is right-angled, the sum of the squares of the shorter sides will equal the square of the hypotenuse

$a^2 + b^2 = \text{hypotenuse}^2$

e.g. $a^2 + b^2 = \text{hypotenuse}^2$
 $3^2 + 4^2 = 5^2$
 $9 + 16 = 25$

Calculate the hypotenuse



Either of the short sides can be labelled a or b

$a^2 + b^2 = \text{hypotenuse}^2$

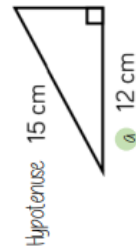
1. Substitute in the values for a and b

$3^2 + 4^2 = \text{hypotenuse}^2$
 $9 + 16 = \text{hypotenuse}^2$
 $25 = \text{hypotenuse}^2$

2. To find the hypotenuse square root the sum of the squares of the shorter sides

$\sqrt{25} = \text{hypotenuse}$
 $5 = \text{hypotenuse}$

Calculate missing sides



Either of the short sides can be labelled a or b

$a^2 + b^2 = \text{hypotenuse}^2$

$12^2 + b^2 = 15^2$

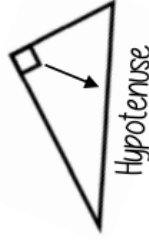
1. Substitute in the values you are given

$144 + b^2 = 225$

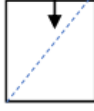
Rearrange the equation by subtracting the shorter square from the hypotenuse squared

Square root to find the length of the side
 $b^2 = 81$
 $b = \sqrt{81} = 9$

Identify the hypotenuse

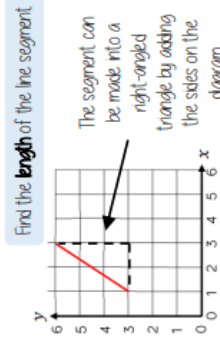


The hypotenuse is always the longest side on a triangle because it is opposite the biggest angle



Polygons can still have a hypotenuse if it is split up into triangles and opposite a right angle

Pythagoras' theorem on a coordinate axis



The line segment is the hypotenuse

$a^2 + b^2 = \text{hypotenuse}^2$

The lengths of a and b are the sides of the triangle

Be careful to check the scale on the axes

Year 9

Music

Term 4

Year 9: Rock'n'Roll – Part 2

Composing and performing a piece

Practice your bass line on your phone app or keyboard.

C Chord Triads	C	+	E	+	G	+	E	+
F Chord Triads	F	+	A	+	C	+	A	+
G Chord Triads	G	+	B	+	D	+	B	+

3 styles – **Walking** (single notes) **Swing doubles** (play twice with swing rhythm) and **Walking with swing** (country style)

Challenge: Use full arpeggio

C	+	E	+	G	+	A	+
Bb	+	A	+	G	+	E	+

Creating Rock'n'Roll lyrics

Rock'n'roll lyric characteristics -

Rock'n'roll lyrics are also more positive and celebratory than the typical blues' lyrics, which are more downbeat.

Typical subjects:

Romance/crushes
Character stories →
Cars and Freedom
School & teenage life

Chuck Berry



Write lyrics for your rock'n'roll song that fits the 12 bars structure.

Challenge: Sing them.

Applying the blues progression
Rock'n'Roll has the same chord structure as the blues.

Practice your bass line with the full blues chord structure.

12-Bar Blues Structure – Counting to know where you are.

I (C)	I (C)	I (C)	I (C)
1,2,3,4	2,2,3,4	3,2,3,4	4,2,3,4
IV (F)	IV (F)	I (C)	I (C)
1,2,3,4	2,2,3,4	3,2,3,4	4,2,3,4
V (G)	IV (F)	I (C)	V (G)
1,2,3,4	2,2,3,4	3,2,3,4	4,2,3,4

Challenge: Use full arpeggio with syncopation.

C	C (bve)	E	C (bve)	G	C (bve)	A	C (bve)
Bb	C (bve)	A	C (bve)	G	C (bve)	E	C (bve)

Year 9: Off beat/Syncopation

If the rhythm always sits on the main beats, it can sound heavy, drudging and dull – To give it a lift, many styles of music use off-beats (**syncopation**) to lift the music and give it a groove you can dance to.

One of the most famous examples is **Reggae** from Jamaica with its heavy bass and choppy bright guitar on the off-beats.



Play the rhythm by saying the count and clapping or tapping on the 'ands'.

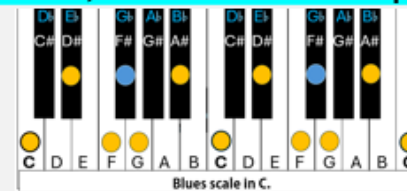
Challenge: Also tap your foot on the main beats.

Applying lead parts

Try recording your bass-line as audio to your phone so you can play it back to play lead riffs on top of it.

Unlike the blues, rock'n'roll lead parts are more energetic and livelier than the blues' which is typically sad and soulful.

Create some rock'n'roll riffs from the blues scale, but make them sound upbeat.



Weak and strong beats

For a more **subtle flowing groove**, instead of playing rhythm like a **switch** (either the note is played or it is not) we can create a **syncopated** feel by **stressing** some beats and playing others **quieter**.

Play the rhythm below – but make the notes with little up arrows (accents) louder, and the others quieter.



Challenge: Make up your own rhythm with accents.

Year 9

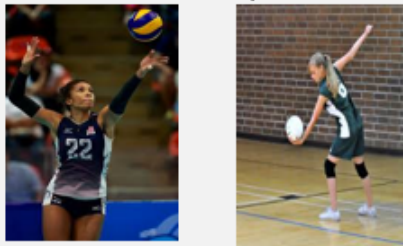
Physical Education

Term 4

7

Serve and Rally Play

Learn the **underarm and overarm serve**.
Keep the ball in play by working as a team—**3 touches per side** (dig-set-spike) is the ideal sequence.



Volleyball

8

Key Rules

Volleyball is played **6 vs 6** (indoors)
Teams rotate **clockwise** each time they win a point on their opponent's serve.

A player **cannot hit the ball twice in a row**.

Key rules include: **no catching, lifting, or throwing the ball, no net touches, and no crossing the centre line.**

Volleyball

9

Lateral Pass

A **lateral pass** (also known as a sideways or backward pass) is the basic passing technique in rugby. Players are **not allowed to pass the ball forward**—it must go **sideways or backwards** to a teammate.



Rugby

10

Tackling or Tagging

In **tag rugby**, remove the tag cleanly and hold it up. In **contact rugby**, tackle safely by lowering your height and wrapping your arms around the legs. Always **keep your head to the side** and follow the safety rules.



Rugby

11

Rucking & Support Play

After a tackle, players must **release the ball**. Teammates form a **ruck** over the ball to retain possession. Support players should run close to provide **passing options** and recycle the ball quickly.



Rugby

12

Offside Rule in Rugby

A key rule is Offside:

A player is **offside** in rugby if they are in **front of the ball** or in front of the teammate who last played the ball. Being offside means you are **not allowed to interfere with play**—you can't tackle, block, or try to win the ball.



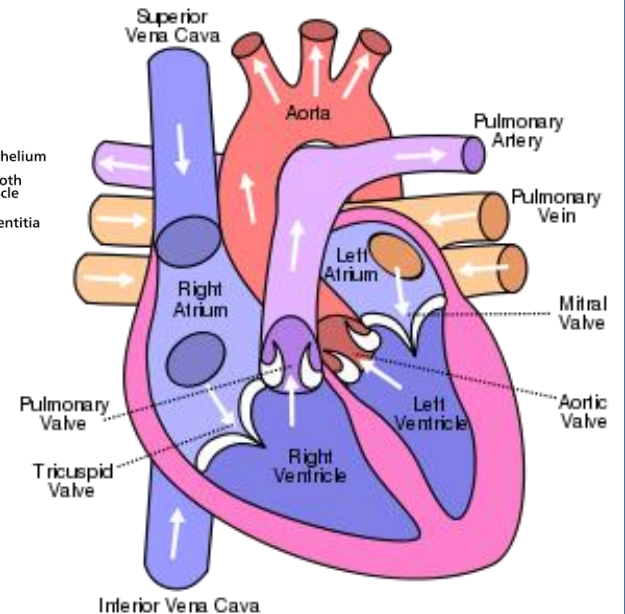
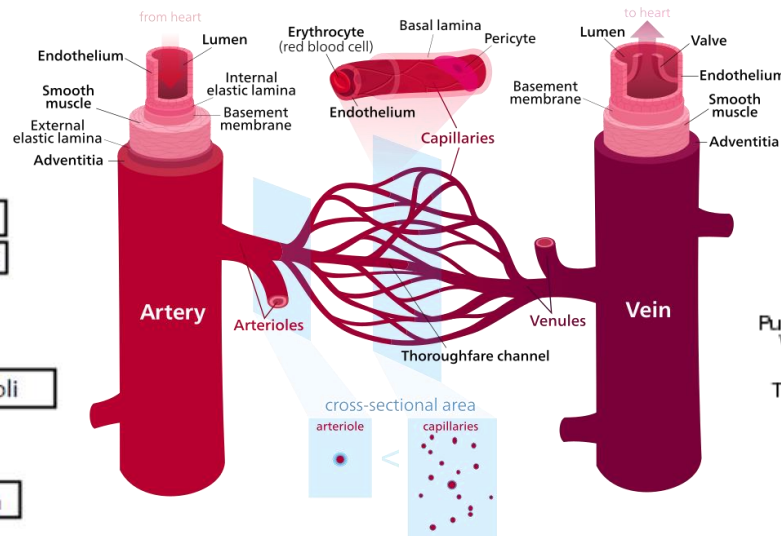
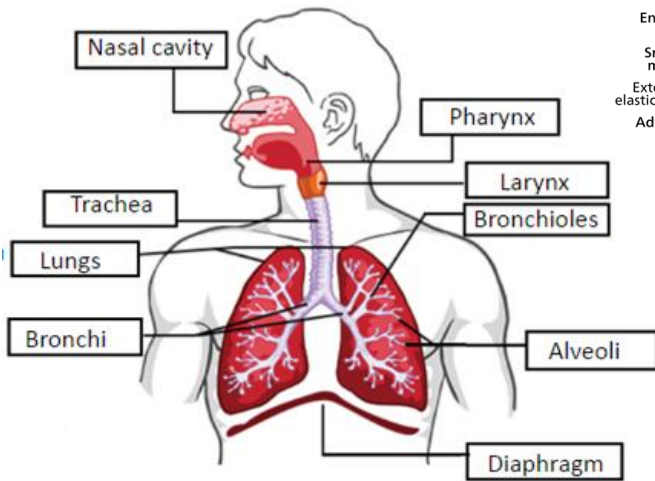
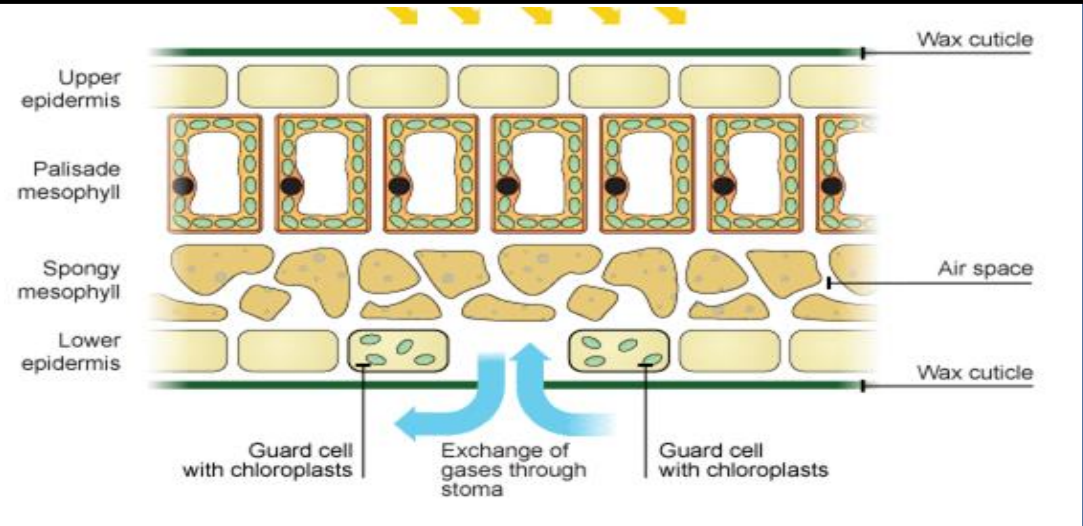
Rugby

Biology

Keywords

Cross section of a leaf

Arteries	Blood vessels that carry blood away from the heart.
Veins	Blood vessels that carry blood to the heart.
Capillaries	The smallest blood vessels where exchange occurs.
Xylem	Transports water around the plant.
Phloem	Carries dissolved sugars around the plant.
Translocation	The movement of sugars from the leaves to the rest of the plant through the phloem.
Transpiration	The loss of water vapour from the leaves of plants through the stomata.



Chemistry

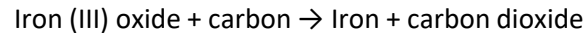
Keywords

Ion	The charged particle formed when an atoms gains or loses electrons
Displacement Reaction	When a more reactive substance takes the place of a less reactive substance
Oxidation	The gaining of oxygen OR the loss of electrons (OIL)
Reduction	The loss of oxygen OR the gain of electrons (RIG)
Base	Ionic compounds that can neutralise acids
Alkali	Soluble bases

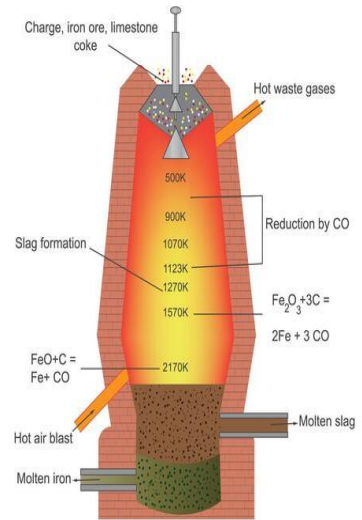
Extracting Metals

Least reactive metals occur native. More reactive metals occur as ores.
For ores containing metals less reactive than carbon, carbon can be used to reduce the metal in a displacement reaction.

Example:



Some less reactive metals can be reduced using hydrogen:



Common Salts

Hydrochloric acid (HCl) → chlorides (Cl⁻)

Sulphuric acid (H₂SO₄) → sulphates (SO₄²⁻)

Nitric acid (HNO₃) → nitrates (NO₃⁻)

General Word Equations

- Metal + oxygen → Metal oxide
- Metal + water → Metal hydroxide + hydrogen
- Metal + acid → Metal salt + hydrogen
- Acid + base → Salt + water
- Acid + alkali → Salt + water
- Acid + carbonate → Salt + water + carbon dioxide

Charges on +ve ions

- Group 1 metals = +1
- Ammonium (NH₄⁺) = +1
- Group 2 metals = +2
- Aluminium = +3

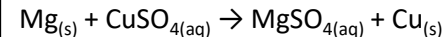
Charges on -ve ions

- Group 7 = -1
- Nitrate (NO₃⁻) = -1
- Hydroxide (OH⁻) = -1
- Sulphate (SO₄²⁻) = -2
- Carbonate (CO₃²⁻) = -2
- Oxide (O²⁻) = -2

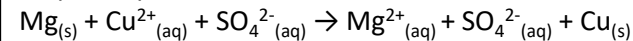
potassium	most reactive	K
sodium		Na
calcium		Ca
magnesium		Mg
aluminium		Al
carbon		C
zinc		Zn
iron		Fe
tin		Sn
lead		Pb
hydrogen		H
copper		Cu
silver		Ag
gold		Au
platinum	least reactive	Pt

Ionic and Half Equations

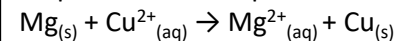
Step 1: Write a balanced symbol equation:



Step 2: Separate the ions:



Step 3: Delete the species that appear exactly the same on both sides:



Half equations:



State symbols: (s) = solid, (l) = liquid, (g) = gas, (aq) = aqueous

Chemistry

Keywords

Hydrocarbon	A compound made of only hydrogen and carbon atoms.
Fractional Distillation	Separating a liquid mixture into fractions differing in boiling point by distillation.
Fuel	Any substance that releases energy when burned.
Cracking	The process by which complex, long chain hydrocarbons are broken down into smaller, simpler ones.
Alkane	A saturated hydrocarbon.
Alkene	An unsaturated hydrocarbon that contains a double bond.
Complete Combustion	When a fuel is burnt in a plentiful supply of oxygen.
Incomplete Combustion	When a fuel is burnt in a limited supply of oxygen.
Condensing	When a gas cools and turns into a liquid.

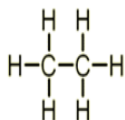
Alkanes

Saturated hydrocarbons

Do not contain a double bond.

General formula C_nH_{2n+2}

Ethane C_2H_6



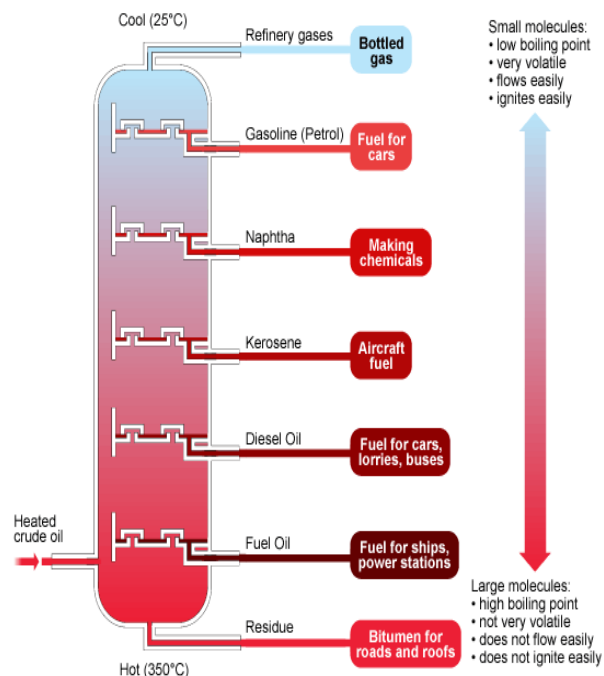
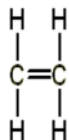
Alkenes

Unsaturated hydrocarbons.

Contain a double bond

General formula C_nH_{2n}

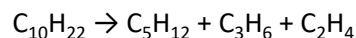
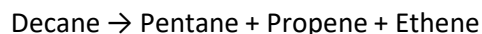
Ethene C_2H_4



Cracking

Cracking is the process of breaking down large hydrocarbons into smaller ones. It uses either a catalyst or steam to make the hydrocarbons thermally decompose. Alkanes and alkenes are the products of cracking.

For example, decane can be broken down with a catalyst at 500°C to make pentane which can be used in petrol



Fractional Distillation

Crude oil is a mixture of different alkanes. Each compound has a different boiling points.

To separate crude oil, it is heated, vapourised and passed through a fractionating column.

In this column, the hydrocarbons cool and condense at different points depending on their boiling points.

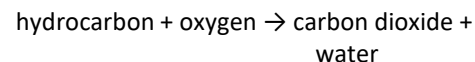
Short chain hydrocarbons condense at the top where it is cool and long chain hydrocarbons condense at the bottom where it is hot.

The mixtures taken out of the column are called fractions

Combustion

During combustion, hydrogen and carbon atoms in fuels are oxidised. There are two types.

Complete Combustion:



Incomplete Combustion:

